Boiling Point Notes

Property of Matter:

- : The temperature in which a liquid turns to a gas
- Factors that affect boiling point (BP):
 - <u>Pressure</u>: **Elevation:** __ pressure = __ BP _ elevation = __ BP
 - # of molecules (salt, sugar, etc):
 - molecules = BP
 - Type of Water vs oil

Separation Technique:

- A mixture of two or more substances that are both can be separated if they have different boiling points.
- **Evaporation:**

Distillation:

Real World Examples:

- ____ (Purify Salt Water) Separation of
- **Treatment Plant**

Property of Matter:

- Boiling Point: The temperature in which a liquid turns to a gas
- Factors that affect boiling point (BP):
 - Pressure:
 pressure = BP
 Elevation:
 elevation = BP
 - # of molecules (salt, sugar, etc):molecules = BP
 - Type of Solvent:
 Water vs oil



Separation Technique:

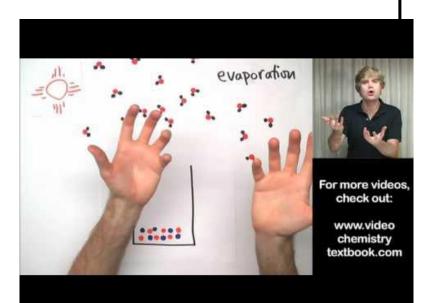
 A mixture of two or more substances that are both soluble can be separated if they have different boiling points.

Evaporation:

 Removes liquids from a mixture to obtain a solid.

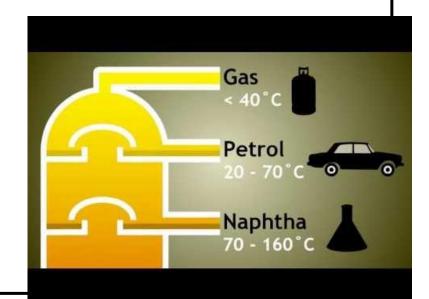
Distillation:

 Heating a mixture to collect the liquid (vapor).

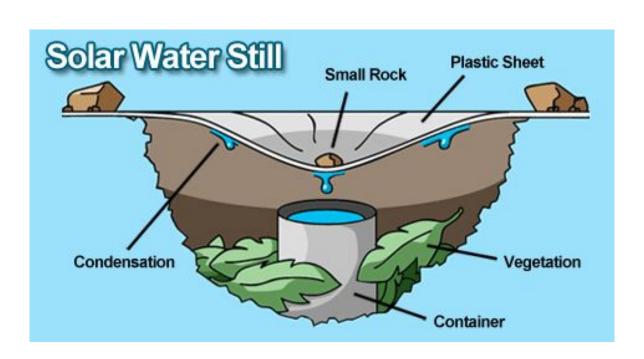


Real World Examples:

- Desalination (Purify Salt Water)
- Separation of Crude Oil
- Water Treatment Plant



Boiling Point Lab:Page 7 and Yellow Sheet



Boiling Point Lab: While mixture is heating...





If we need to wait longer...

Boiling Point Lab Clean up Procedures:

- Finish your analysis questions first!
- Wait until everything is cooled enough to touch
- Leave hot plate and hot pads at your table group
- Wipe up any salt or water on the desks
- Dump your mixture and place aluminum tin and plastic supply bin back on the table just as you found it