Name:	Date:
Directions: leftovers' sir	go to pHet online. Click on 'chemistry sims'. Click on 'reactant, products and n.
Or try -	Reactants, Products and Leftovers
On 'sandwi	ches' ' cheese' sim:
1. Start	with 3 breads and 2 cheese.
a	. How many products are made?
b	. What are the leftovers?
2. Add	a slice of cheese.
а	. How many products are made?
t	. What are the leftovers?
3. Now	I have 8 breads and 3 cheese.
a	. How many products are made?
b	. What are the leftovers?
4. Now	try 8 breads and 8 cheese.
a	. How many products are made?
b	. What are the leftovers?
	the 8 breads and 8 cheese – what did you run out of first? (this is your limiting
reac	ant)
	play with it and find out the max number of products which can be made without eftovers.
a	. How many of each reactant were used?
b	. How many of the products were produced?

Name	:	Date:
On 'sa	andwic	hes meat and cheese' sim:
7.	Play w	vith the sim and try to make only one sandwich with no leftovers.
	a.	How many of each reactant were used?
8.	Now n	nax every reactant out.
	a.	How many of each reactant is that?
	b.	How many products can be made?
	C.	How many of these leftovers are created?
	d.	Which reactant was your limiting reagent?
9.	Now n	naximize the number of products without having any leftovers.
	a.	How many of each reactant are used?
	b.	How many products were made?
On 'sa	andwic	hes custom' sim
10	. Set it o	on a triple decker meat and cheese sandwich at the top.
	a.	How many of each reactant will you need to make one triple decker meat and
		cheese sandwich?
11	. Now n	nax out how many products can be made without leftovers.
	a.	How many of each reactant did you use?
	b.	How many products did you produce?
12	. Now n	nax out all of the reactants.
	a.	How many products were produced?
	b.	How many of each leftover were there?
13	s. What i	is the limiting reactant in the triple decker sandwich?

Name:	Date:				
Using the 'molecules – make water' sim					
14. Make a single water molecule.	14. Make a single water molecule.				
a. How many of each reactant were used?					
15. a. What is the mole ratio (how many of each reactant a	and product are there)?				
b. Now write it as a chemical reaction:					
16. Max out how many water molecules that can be made	without leftovers/excess.				
a. How many of each reactant were used?					
b. How many products were produced?					
c. What is the mol ratio (how many of each reacta	ant and product are there)?				
d. Now express it as a chemical equation:					
17. Now max out every reactant.					
a. How many products were made?					
b. How many and what were the leftovers/excess	?				
c. What is the limiting reactant?					
Using the 'molecules – make ammonia' sim					
18. Make a single ammonia molecule.					
a. How many of each reactant were used?					
b. What is the mol ratio?					

Name:	Date:
C.	Write it as a chemical equation:
19. Play w	rith it until you find the max amount of ammonia that can be produced without any
leftove	
a.	How many of each reactant are used?
b.	How many ammonia molecules are produced?
C.	Write it as a chemical equation:
20. Now n	nax all of the reactants.
a.	How many ammonia molecules are produced?
b.	How many of each leftover are there?
C.	What is the limiting reactant in this reaction?
Using the 'm	olecules – combust methane' sim
21. Produ	ice a single carbon dioxide molecule without any leftovers.
a.	How many of each reactant are used?
b.	How many of each product are produced?
C.	What is the mole ratio:
d.	Write the reactant in a chemical equation form:
22. Now n	nax all of the reactants out.
a.	How many of each product are formed?

Name:		Date:
	b. How many of what is leftover are there?	
23. Nov	w maximize the number of products produced without leftovers.	
	a. How many of each reactant are consumed?	
	b. How many of each product are produced?	
	c. What is the mol ratio?	
	d. Now write it as a chemical equation:	
24. a. V	What is the limiting reagent in this reaction?	
b. H	low do you know?	
Now use th	e 'Reactants, Products & Leftovers – Game'	
	over when you have scored an 8 or better on level 1for this assignment.	(teacher initial) You're
	over when you have scored an 8 or better on level 2up to a 'B' for this assignment.	_ (teacher initial)
	over when you have scored an 8 or better on level 3up to an 'A' for this assignment	_(Teacher initial)