

All about that base

A Bm E

You know I'm all about that base, about that base, about that base,
no H^+

I'm all about that base, about that base, no H^+

I'm all about that base, about that base, no H^+

I'm all about that base, about that base

You know the pH scale I kind of like it, what do those #'s mean?

0-14

Power of Hydrogen, or the H 's

so all the acids are under pH 7

What about the bases, full of hydroxide

put it with an acid, neutralization

In base we don't find H , only OH

So all the bases are over pH7

Where can I find a base? Just go look for some soap

Cause soap is slippery, bitter, and made from some OH 's (OH 's)

And if you ever find, some Sodium Hydroxide take caution

this base will liquify your phospholipid bilayer into soap.

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Naming Acids

H with single anion (negatively charged ion)- Just "Hydro-" and name of element with "-ic" for one word. Then add acid at the end.



H with polyatomic ion- Do not use hydro at the start of the name. Just say the name of the polyatomic ion and change the ending. "-ate" turns into "-ic", and "-ite" turns into "-ous" then add acid to the end of the name.

