

Name _____

H. Chem, Period _____

Ionic Nomenclature WS #2

Name of Compound	Positive Ion	Negative Ion	Formula
sodium iodide			
silver sulfide			
barium nitrate			
aluminum acetate			
strontium nitride			
lithium phosphate			
potassium bromide			
silver sulfite			
chromium (III) chloride			
iron (II) phosphide			
lead (IV) carbonate			
manganese (IV) oxide			

Formula	Positive Ion	Negative Ion	Compound Name
Fe_2S_3			
Cu_3N_2			
$\text{Sr}(\text{NO}_3)_2$			
Mg_3P_2			
FeN			
CaO			
$\text{Ba}(\text{C}_2\text{H}_3\text{O}_2)_2$			
Li_2CrO_4			
SnCl_2			
$\text{Al}_2(\text{SO}_3)_3$			
$\text{Zn}_3(\text{PO}_4)_2$			
Na_2Se			