



# РАСТВОРЕНИЕ МЕТАЛЛОВ В КИСЛОТЕ

- $\text{HCl} + \text{Fe} =$
  - $\text{H}_2\text{SO}_4(\text{p-p}) + \text{Al} =$
  - $\text{HBr} + \text{Ca} =$
  - $\text{HF} + \text{Na} =$
  - $\text{HNO}_3(\text{конц}) + \text{Ag} =$
  - $\text{H}_3\text{PO}_4 + \text{Ba} =$
  - $\text{HI} + \text{Cr} =$
  - $\text{HNO}_3(\text{p-p}) + \text{Cu} =$
  - $\text{CH}_3\text{COOH} + \text{Mg} =$
  - $\text{H}_2\text{SO}_4(\text{конц}) + \text{Cr} =$
  - $\text{HCl} + \text{Al} =$
  - $\text{H}_2\text{SO}_4(\text{p-p}) + \text{Zn} =$
  - $\text{HBr} + \text{Ni} =$
  - $\text{HF} + \text{Ba} =$
  - $\text{HNO}_3(\text{конц}) + \text{Ba} =$
  - $\text{H}_3\text{PO}_4 + \text{K} =$
  - $\text{HI} + \text{Fe} =$
  - $\text{HNO}_3(\text{p-p}) + \text{Mg} =$
  - $\text{CH}_3\text{COOH} + \text{Al} =$
  - $\text{H}_2\text{SO}_4(\text{конц}) + \text{Fe} =$
- 