


# Storyline Unit Design

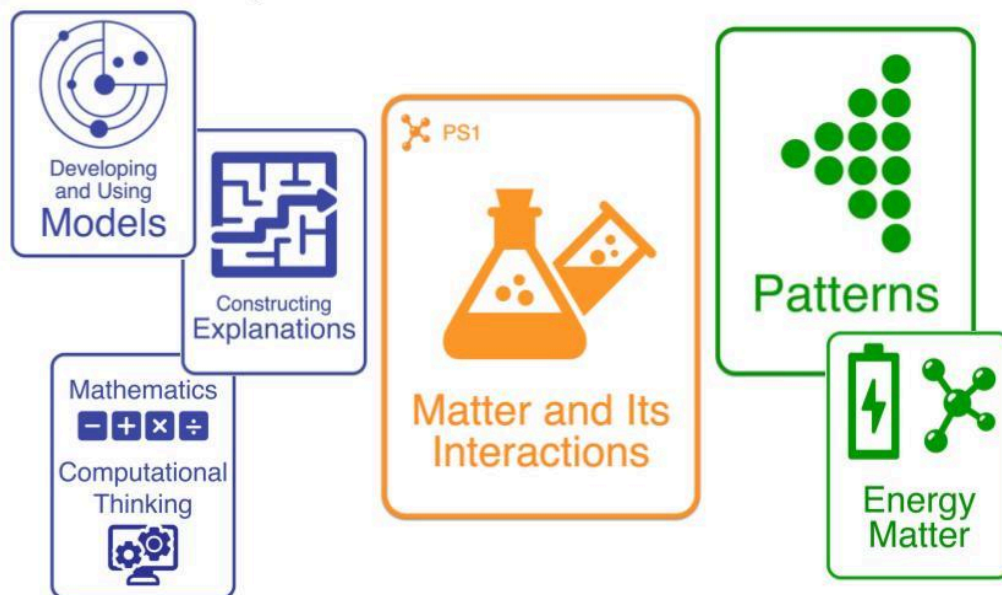
## Understanding by Design (UbD) Template\*

Unit		Course(s)	
Designed by		Time Frame	

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### Anchor Model

## Atoms, Elements and Molecules



## Stage 1: Desired Results

### Performance Expectations

#### **HS-PS1-1: Valence Electrons and Properties of Elements**

Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level of atoms.

#### **HS-PS1-2: Simple Chemical Reactions**

Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

#### **HS-PS1-7: Conservation of Atoms in Chemical Reactions**

Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction. (Energy and Matter)

### Anchoring Phenomenon

[Anchoring Phenomenon Worksheet](#)

### Enduring Understandings

### Essential Questions

## Stage 2: Assessments

HS-PS1-1 - [Ordering Reactivity](#)  
HS-PS1-2 - [Predicting Products](#)  
HS-PS1-7 - [Toe Warming Chemistry](#)

[Assessment Screening Tool Slides](#)





### Backward Design Elements

What new skills (practices) will students need to learn?

What thinking concepts will students need to learn?

What science concepts will students need to learn?

## Stage 3: Learning Plan

 Phenomenon or Problem	 Learning Performance - What will they do?  The three dimensions woven together into a single learning performance.	 Why is this important?  How does this activity help build understanding of the anchoring phenomenon.	 Learning Experience - How will they do it?  Graphic organizers, protocols, scaffolds, labs, mini-lesson, student discourse, etc.
Design a Fireworks display <a href="#">Fireworks accident</a>	Students will <b>ask questions</b> the <b>cause and effect</b> of using different <b>elements and chemical reactions</b> to create a fireworks display	All students have experienced <b>elements and chemical reactions</b> from fireworks.	Students will read an article and watch a video about the fireworks accident and fill out a <b>graphic organizer</b> about their <b>questions</b>
Formative Assessment - What information are you collecting to know that they met the target?			
Which elements make specific colors in fireworks?	Students will <b>obtain, evaluate, and communicate information</b> by <b>planning and carrying out investigations</b> burning wood splints soaked in different salt solutions and discover <b>patterns</b> of elements and the colors they produced when burned <b>based on the patterns of electrons in the outermost energy level of atoms.</b>		Students will design and plan an investigation to test many different salts (some with the same metallic salt). They will reflect on the patterns that they see with the colors and the elements in the solutions.
Formative Assessment - What information are you collecting to know that they met the target?			
<a href="#">Fireworks accident</a>	Students will <b>design solutions to support the claim that atoms, and therefore mass, are conserved during a chemical reaction</b> and create a plan for a fireworks display		
Formative Assessment - What information are you collecting to know that they met the target?			
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## Materials / Resources

### **Vocabulary**

#### HS-PS1-1

Atoms  
Nucleus (positively-charged)  
- Protons  
- Neutrons  
Electrons (negatively-charged)  
Valence electrons  
Periodic table  
Elements  
Elemental properties (e.g. reactivity of metals, types of bonds formed, number of bonds formed, reactions with oxygen)  
Reactivity  
Patterns

#### HS-PS1-2

Chemical reaction (e.g. sodium and chloride, carbon and oxygen, carbon and hydrogen)  
Reactants and products  
Bonds (i.e. ionic, covalent)  
Reaction type (e.g. formation of ionic compound, combustion of hydrocarbons)  
Electronegativity  
Main group elements  
Valence electrons  
Periodic table  
Chemical properties  
Patterns

#### HS-PS1-7

Chemical reaction  
Balanced chemical equation  
Reactants and products  
Atoms (or molecules or ions)  
Moles  
Mass  
Conservation of mass  
Atomic scale  
Macroscopic scale  
Matter

### **Mini Lessons**

Patterns Level 7 - [Patterns Level 7 - Causal Patterns at Varying Scale](#)

Patterns Level 7 - [Patterns Level 7 - Causal Patterns at Varying Scale Thinking Slides](#)

### **Graphic Organizers**

[Phenomena Observation Graphic Organizer](#)

[Questioning Graphic Organizer](#)

[Modeling Graphic Organizer](#)

[Planning an Investigation Organizer - Experimental](#)

[Planning an Investigation Organizer - Observational](#)

[Investigation Evidence Organizer](#)

[Engaging in Argumentation Organizer](#)

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