

Name \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

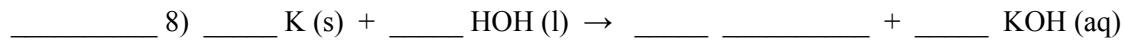
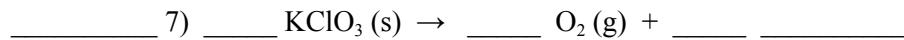
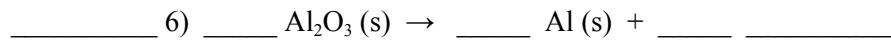
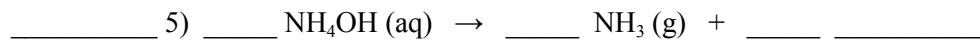
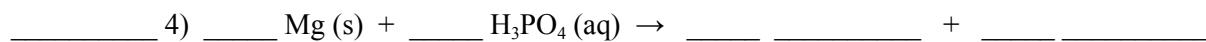
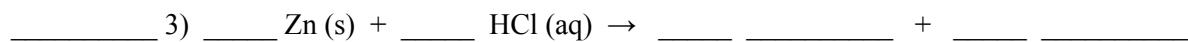
## Chemistry U09 - 08      Activity 4 -7 Practice Drill: Chemical Equations

Complete and balance the following equations for chemical reactions. In the space at the left of each equation, classify the category of the reaction.

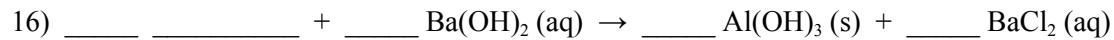
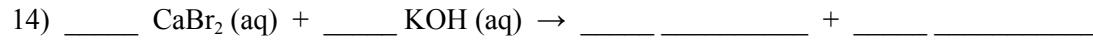
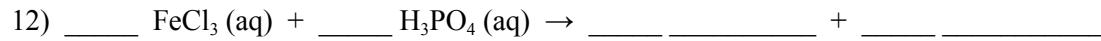
Use    SYN - synthesis DR - double replacement

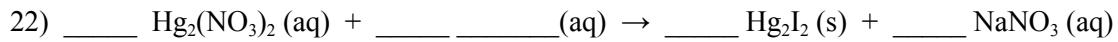
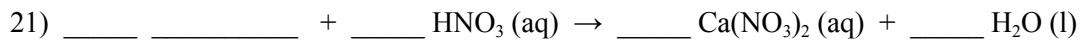
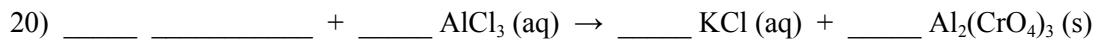
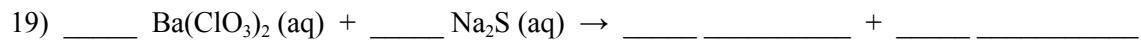
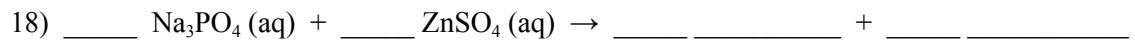
C - combustion      SR - single replacement

D - decomposition

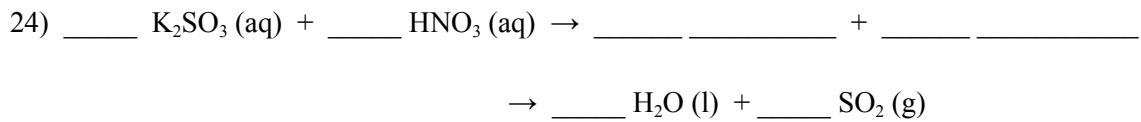
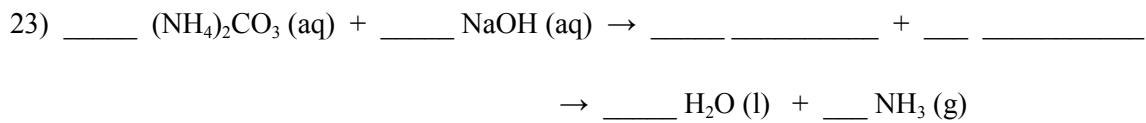


Complete and balance the following equations for chemical reactions.



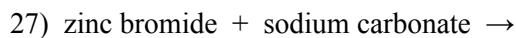
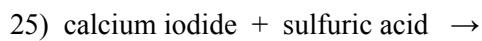


**Sometimes a product will spontaneously decompose into two more products.**

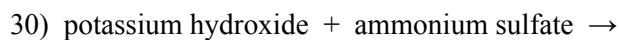


To which category do the reactions 11 - 24 belong? \_\_\_\_\_

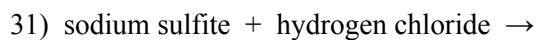
Write balanced chemical equations for each of the following double replacement reactions.



(one product decomposes further into two more products)



(one product decomposes further into two more products)



(one product decomposes further into two more products)