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Students will develop models to describe the atomic composition of simple molecules and extended structures. VOCABULARY: Molecule- LESSON: Use the reading packet to fill in the following notes.			
		Covalent A is the force of attraction that holds together This forms a Forms only between (These are found on the right side of the table). Draw Figure 1. Circle the electrons that are being shared.	An ionic bond is the force of attraction that holds together positive and negative This happens between and Sodium loses an electron and becomes an ion with acharge (symbol+) Chlorine gains an electron and becomes an with a negative charge (symbol)
		Covalent bonds form because they give atoms a more Arrangement of electrons.	Opposite electrical chargeseach other.