

Name: _____

Copper Chloride

Information

Copper (I) chloride:

Melting point = 426°C

Conductivity as solid = nil

Explain the properties of CuCl shown above in terms of structure and bonding.

This image shows a single sheet of white paper with horizontal ruling lines. The lines are evenly spaced and run across the width of the page. There are no margins, text, or other markings on the paper.