

Question 4.1:

Given data:

Partial pressure of oxygen in alveoli = $1.4 \times 10^{-4} \text{ M} = 1.4 \times 10^{-4} \text{ mol/L} = 1.4 \times 10^{-4} \times 10^3 \text{ mol/m}^3 = 1.4 \times 10^{-1} \text{ mol/m}^3$

Partial pressure of oxygen in blood = $5.6 \times 10^{-5} \text{ M} = 5.6 \times 10^{-5} \text{ mol/L} = 5.6 \times 10^{-5} \times 10^3 \text{ mol/m}^3 = 5.6 \times 10^{-2} \text{ mol/m}^3$

Thickness of membrane = $20 \text{ } \mu\text{m} = 20 \times 10^{-6} \text{ m}$

$D_{O_2} = 1 \text{ } \mu\text{m}^2/\text{msec} = 10^{-6} \text{ m}^2/\text{sec} = 10^{-3} \text{ m}^2/\text{sec}$

To find:

Rate of oxygen transport = ?

Formula:

Rate of oxygen transport = $\frac{DO_2}{\Delta x} (P_{O_2, \text{alveoli}} - P_{O_2, \text{blood}})$

Calculations:

By putting values:

Rate of oxygen transport = $\frac{10^{-3} \text{ m}^2/\text{sec}}{20 \times 10^{-6} \text{ m}} (1.4 \times 10^{-1} \text{ mol/m}^3 - 5.6 \times 10^{-2} \text{ mol/m}^3)$

Rate of oxygen transport = $10^{-6.3}/20 (0.084 \text{ mol/m}^3) = 4.2 \text{ mol/s per m}^2$

Question 4.2:

Given data:

Partial pressure of oxygen in alveoli = $1.4 \times 10^{-4} \text{ M} = 1.4 \times 10^{-4} \text{ mol/L} = 1.4 \times 10^{-4} \times 10^3 \text{ mol/m}^3 = 1.4 \times 10^{-1} \text{ mol/m}^3$

Partial pressure of oxygen in blood = $5.6 \times 10^{-5} \text{ M} = 5.6 \times 10^{-5} \text{ mol/L} = 5.6 \times 10^{-5} \times 10^3 \text{ mol/m}^3 = 5.6 \times 10^{-2} \text{ mol/m}^3$

Thickness of membrane = $20 \text{ } \mu\text{m} = 20 \times 10^{-6} \text{ m}$

$$D_{O_2}=0.5 \mu\text{m}^2/\text{msec}=0.5 \times 10^{-6} \text{ m}^2/\text{sec}=0.5 \times 10^{-3} \text{ m}^2/\text{sec}$$

To find:

Rate of oxygen transport=?

Formula:

$$\text{Rate of oxygen transport} = \frac{DO_2}{\Delta x} (P_{O_2}.\text{alveoli} - P_{O_2}.\text{blood})$$

Calculations:

By putting values:

$$\text{Rate of oxygen transport} = \frac{0.5 \times 10^{-3} \text{ m}^2/\text{sec}}{20 \times 10^{-6} \text{ m}} (1.4 \times 10^{-1} \text{ mol/m}^3 - 5.6 \times 10^{-2} \text{ mol/m}^3)$$

$$\text{Rate of oxygen transport} = 0.5 \times 10^{-3} / 20 (0.084 \text{ mol/m}^3) = \mathbf{2.1 \text{ mol/s per m}^2}$$

Question 4.3

Rate of oxygen transport per square meter through injured lung is 2.1 mol/s. Rate of oxygen transport per square meter through healthy lung is 4.2 mol/s. considering both values, it is inferred that rate of oxygen transport is reduced to half in injured tissue.

Question 4.4

Given data:

$$\text{Partial pressure of oxygen in alveoli} = 1.4 \times 10^{-4} \text{ M} = 1.4 \times 10^{-4} \text{ mol/L} = 1.4 \times 10^{-4} \times 10^3 \text{ mol/m}^3 =$$

$$1.4 \times 10^{-1} \text{ mol/m}^3$$

$$\text{Rate of oxygen transport} = 4.2 \text{ mol/s per m}^2$$

$$\text{Thickness of membrane} = 20 \mu\text{m} = 20 \times 10^{-6} \text{ m}$$

$$D_{O_2} \text{ of half injured lung} = 0.5 \times 1 \mu\text{m}^2/\text{msec} = 0.5 \times 10^{-6} \text{ m}^2/\text{sec} = 0.5 \times 10^{-3} \text{ m}^2/\text{sec}$$

$$\text{Since } D_{O_2} \text{ is directly proportional to area, quarter injured tissues' } D_{O_2} \text{ value} = 0.75 \times 10^{-3} \text{ m}^2/\text{sec}$$

To find:

Partial pressure of oxygen in blood=?

Formula:

$$\text{Rate of oxygen transport} = \frac{DO_2}{\Delta x} (P_{O_2.\text{alveoli}} - P_{O_2.\text{blood}})$$

Calculations:

By putting values:

$$4.2 \text{ mol/s/m}^2 = \frac{0.75 \times 10^{-3} \text{ m}^2/\text{sec}}{20 \times 10^{-6} \text{ m}} (1.4 \times 10^{-1} \text{ mol/m}^3 - P_{O_2.\text{blood}})$$

$$8.4 \times 10^{-5} / 0.75 \times 10^{-3} = 1.4 \times 10^{-1} \text{ mol/m}^3 - P_{O_2.\text{blood}}$$

$$0.112 = 1.4 \times 10^{-1} \text{ mol/m}^3 - P_{O_2.\text{blood}}$$

$$P_{O_2.\text{blood}} = 1.4 \times 10^{-1} - 0.112 = 0.14 - 0.112 = 0.028 \text{ mol/m}^3$$

Value of partial pressure of oxygen in blood should be **0.028 mol/m³ = 20000 mmHg**