Name: Date: Color:

Osmosis in Gummy Bears Lab



Background Information:

The term **[osmosis]** refers to the movement of water molecules, usually across a membrane. The molecules move from a region of **[high]** concentration to **[low]** concentration.

Cells, like other living things have to do certain things to survive. Two examples are acquiring **[food, water]** and removing **[waste]**. These actions use chemical reactions that require water. In plants, one chemical reaction that uses water is called **[photosynthesis]**.

In plants, water is stored in vacuoles. The rigid cell **[wall]** prevents the cell from bursting. Some protists have an organelle called a **[contractile vacuole]** that squeezes the water out. Many animal **[cells]** have no good way for regulating water. They swell up when too much water moves in, and can shrivel up if the cell loses water.

Today, we will **[use gummy bears]** to model the cell, and see how they manage environments with different sugar **[concentrations]**. We will place the gummy bears in 2 different baths. One cup will have a liquid with a high sugar concentration and the other cup will have a liquid with a low sugar concentration. We will measure the length, width, and height of each gummy bear to determine its approximate **[volume]**.

Identify the variables:

- Independent variable: sugar concentration
 (what is different about the environment that the two gummy bears are in?)
- 2. Dependent variable: **volume of the gummy bear** (what data are you collecting?)

Name	:	Date:	Color:
(How a. b. c.	f Constants: will you keep both gummy bear treatr cups are the same same brand of gummy bear (ingre temperature amount of liquid	·	u may add more than A-D):
1.	theses: If the gummy bear is placed in the If the gummy bear is placed in the		
I thin	oonse: k my first hypothesis will be prover k my second hypothesis will be pro		

Materials:

- 2 small plastic cups
- 2 different-colored gummy bears
- liquid with high sugar concentration
- liquid with low sugar concentration
- sharpies
- metric ruler
- plastic spoon (day 2)
- paper towel

Procedure:

1. Collect 2 plastic cups, a ruler, and 2 different-colored gummy bears

- 2. Label one cup high sugar concentration, your class color and initials.
- 3. Label one cup low sugar concentration, your class color and initials.
- 4. Measure the height, width, and length of each gummy bear (they might be different!).

 Record your measurements in the table.



- 5. Pour 10mL of liquid with low sugar concentration in one cup, place one gummy bear in the cup.
- 6. Pour 10mL of liquid with high sugar concentration in one cup, place one gummy bear in the cup.
- 7. Cover both cups with saran wrap
- 8. Place the cups on your color's tray [MAKE SURE YOU LABELLED THEM!]
- 9. Allow the bears to soak overnight.
- 10. **On Day 2**, use a plastic spoon to GENTLY place the bear from the liquid with low sugar concentration on a paper towel.
- 11. Measure the height, width, and length of the gummy bear. Record your measurements in the table.
- 12. Use a plastic spoon to GENTLY place the bear from the liquid with high sugar concentration on a paper towel.
- 13. Measure the height, width, and length of the gummy bear. Record your measurements in the table.

Data Table: Use metric units.

BEFORE	Low Sugar Concentration Bear Color:	High Sugar Concentration Bear color:
length	10 mm	mm
width	mm	mm

	Date:	Color:	
height	mm	mm	
volume	mm^3	mm^3	
	•	•	
AFTER	Low Sugar Concentration Bear	High Sugar Concentration Bear	
length	20 mm		
width			
height			
•			
volume			
volume	= change) (change/before x 100 Low Sugar Concentration Bear	= percent change) High Sugar Concentration Bear	
volume Data Analysis (after before 20-10=10 10/10*100=	Low Sugar Concentration	High Sugar Concentration	
volume Data Analysis (after before 20-10=10 10/10*100= PERCENT CHANGE	Low Sugar Concentration Bear	High Sugar Concentration	
volume Data Analysis (after before 20-10=10 10/10*100= PERCENT CHANGE length	Low Sugar Concentration Bear	High Sugar Concentration	

What change did you notice?

Why do you think this happened?

How does this experiment demonstrate osmosis?

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What could have affected your results?

GUMMY BEAR OSMOSIS LAB RUBRIC

Item	Points	Self-Assess	Earned	Comment
Title Is in the format "Gummy Bear Osmosis by First Last (color)	5			
Hypothesis: Has two parts, is in "if/then" format. Explains prediction.	5			
Introduction Explains the important vocabulary, briefly explains experiment and purpose	10			
Procedure: is numbered and could be replicated. Describes how data is collected	5			
Materials: is bulleted, may be copied from lab sheet	5			
Results/Analysis: Describes the data observed in gummy bears, does not give reasons why yet.	10			
Conclusion: Restates purpose, results, and if hypothesis was proven. Asks further questions.	20			
Data Table: is attached, is neat, shows metric units.	10			
Neatness/Spelling: No funny fonts or emoticons. Follows editing checklist.	5			
Self-assessment on rubric	5			
Total				

Name: Date: Color:

- Does each sentence begin with a capital letter?
- Check all ending punctuation.
- Is every sentence a complete sentence with a noun and a verb?
- Is every word spelled correctly?
- Are all proper nouns capitalized?
- Are any word misused? (it's/its, your/you're, etc.)
- Is all text in the same font style and size?

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