The lab questions listed on this page are the ONLY ONES you must answer

Don't Panic

use the LINEST function on google spreadsheet to find the slope (much easier than excel):

https://support.google.com/drive/answer/3094249

how to type it:

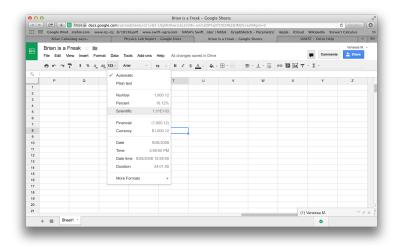
=LINEST(y-values (energy loss),x-values(frequency),1,1)

the value in the top left cell is the slope

make sure to highlight all the data, go to the tab labeled "123", and choose "scientific"

Here is a speadsheet

https://docs.google.com/spreadsheets/d/1dA17oJ2EoEavNWZbv0pUD_fF_2HfgBAfBSkB-CHh 5cQ/edit#gid=0 how to use LINEST.



Outline: https://www.dropbox.com/s/li3vfq99bt5cd0r/Lab%20Report%20%2812th%20Grade%2

0Framework%29.doc

Rubric: https://docs.google.com/file/d/0BzqIpKV_tZjqYnhieXhRd3hMZGs/edit

Phantastic Photon: https://docs.google.com/file/d/0BzqIpKV tZjqTFBIOkVoTVNrcUU/edit

LED: https://docs.google.com/file/d/0B-w7c4065bNuTmpkTU92UkZnUDg/edit

Lab Questions:

1) What has made this experiment meaningful to your science experience during senior year?

(Each person in your group should write a paragraph answer for this question. Clearly label

which answer correspond to group members.)

2) In nanofabrication labs, scientists make microprocessor chips in cleanrooms; these rooms

prevent dust and human hair from interfering with the creation of microprocessor chips. The

chips are very sensitive to light because they are made with layers of chemicals. One chemical

in particular – photoresist - will completely disintegrate if it is exposed to UV light, thereby

destroying the chip's ability to be used in an electronic device.

a. Why are cleanrooms where photoresist is used illuminated with yellow light instead of

standard fluorescent lights?

b. Why does photoresist disintegrate when exposed to UV light?

V to eV:

Refer to this website for an explanation of this (as recommended by Pittman):

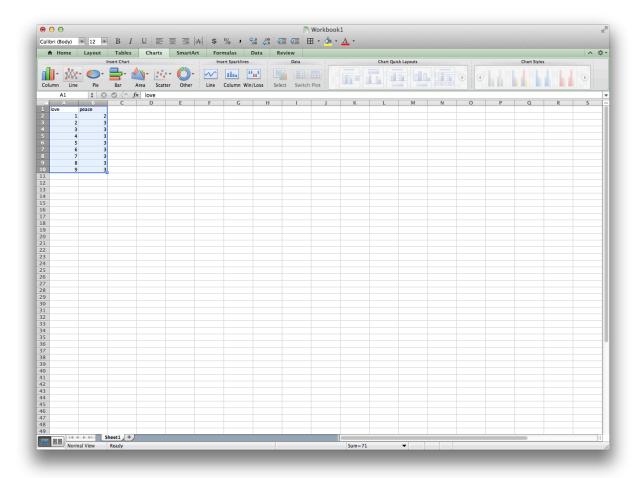
http://www.rapidtables.com/convert/electric/volts-to-ev.htm

Finding Planck's Constant:

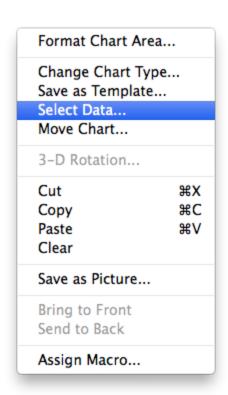
You must make a graph with energy loss (J) on one axis and frequency on the other. To do this,

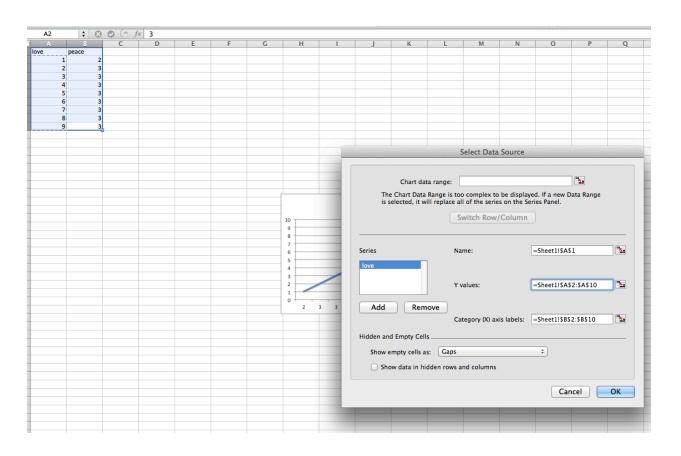
copy and paste the data onto an Excel sheet, make a graph (just go to "charts" and choose the

line graph option).

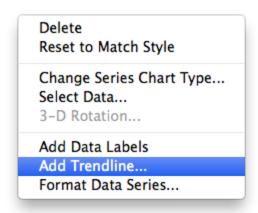


Right click the graph and choose "select data", and highlight the data you want for the x-axis and the y-axis.

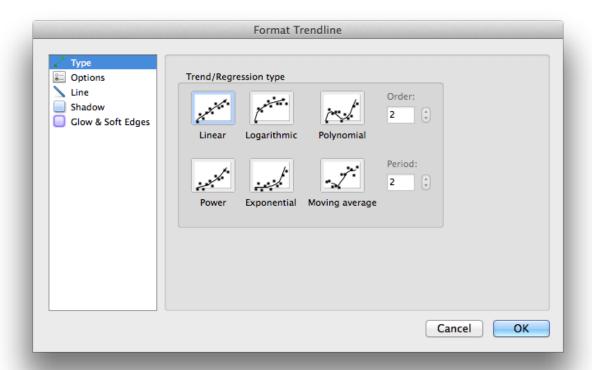




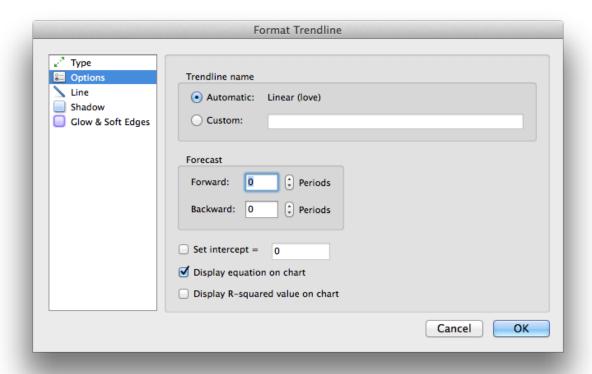
Next, click on the line your graph, right click, and choose "Format Trendline".



Now choose "linear", go to options...



...and check that box next to "Display equation on chart."



There is a way to do this on google spreadsheets, as described at the top of this page.

In fact, it is the better way! Use google spreadsheet! Way easier!