

SCHEME OF WORK

Name of teacher:

Year: 2023

Class/Stream: FORM TWO

Name of School:

Term: 1 & 2

Subject: CHEMISTRY

Competence	General objective	Number of weeks	Week	Main Topic	Sub-topic	Periods	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Remarks
Preparing and testing properties of simple gases.	To explain the preparation and properties of simple gases.	3rd	JANUARY	OXYGEN	1.1 Preparation and properties of Oxygen	4	i) Guiding students on preparation of Oxygen from Hydrogen peroxide ii) Guiding students to carry out the characteristic test for Oxygen gas. iii) Leading a discussion on the physical and chemical properties of oxygen	i) Carrying out an experiment to prepare Oxygen from H_2O_2 or by heating $KClO_3$ ii) Burning metals and some non-metals to oxygen. iii) In groups, to discuss the risks of using $KMnO_4$ and HgO to prepare oxygen in laboratory.	H_2O_2 , $KClO_3$, MnO_2 , flat bottomed flask, beehive shelve, delivery tube, troughs, gas jar, water, thistle funnel, freshly prepared O_2 , P , litmus paper, candle, combustion spoon, Mg ribbon, C , S , Ca granules, wooden splint, $KMnO_4$.	i) Chemistry for secondary schools, form 1&2, Oxford. ii) O-level CHEMISTRY Form 2, BEN. iii) O-level CHEMISTRY Form 2 Interactive CD, BEN. iv) You tube(for video clips)	i) Ability to prepare and collect a sample of oxygen gas in the laboratory. ii) Ability to perform simple experiments on physical and chemical properties of oxygen has. iii) Ability to explain	

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in th e la b or at or y.		4th			1.2 Uses of oxygen	2	i) Guiding students in groups to discuss daily life uses of oxygen. ii) Guiding students to discuss the relationship between uses of oxygen and properties.	i) Discussing about uses of oxygen ii) Discussing about the relationship between uses of oxygen and properties.	Wall charts, flip charts, pictures showing uses and properties of oxygen in the mentioned process.		properties of oxygen. i) Ability to list the uses of oxygen in daily life. ii) Ability to relate some uses of oxygen to its properties.
C o m p et e nc e	Gene ral objec tive	W e ek N o n t h	Mai n Top ic	Sub-topi c	P e ri o d s	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Ren

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<p style="text-align: center; font-size: 1.5em; margin: 0;">F E B R U A R Y</p>	<p style="text-align: center;">4th & 1st</p>	<p style="text-align: center;">HYDR OGEN</p>	<p style="text-align: center;">2.1 Preparation and properties of Hydrogen.</p>	<p style="text-align: center;">4</p>	<p>i) Demonstration on the preparation of a small sample of hydrogen using Zinc and dilute HCl ii) Demonstrating an experiment on 'pop' sound test for hydrogen. iii) Carrying out an experiment on reduction of CuO using H₂ gas. iv) Guiding a discussion on the physical and chemical properties of H₂ gas.</p>	<p>i) Observing the demonstration on the preparation of sample of hydrogen using Zinc and dilute HCl. ii) Demonstrating an experiment on 'pop' sound test for hydrogen. iii) Discussing on the physical and chemical properties of H₂ gas.</p>	<p>Zinc granules, dil. HCl, trough, thistle funnel, beehive shelves, flat bottomed flask, test tube, wooden splints, CuO, PbO, litmus paper, CoCl, paper, combustion tubes, source of heat, anhydrous CaCl₂, wall chart showing physical and chemical properties of H₂.</p>	<p>i) Chemistry for secondary schools, form1&2, Oxford. ii) O-level CHEMISTRY Form 2, BEN. iii) O-level CHEMISTRY Form 2 Interactive CD, BEN. iv) You tube(for video clips)</p>	<p>i) Explain the preparation of a sample H₂ in the laboratory. ii) Ability to explain properties of H₂ gas.</p>
<p style="text-align: center;">1st</p>	<p style="text-align: center;">2</p>	<p style="text-align: center;">2.2 Uses of Hydrogen</p>	<p style="text-align: center;">2</p>	<p>i) Leading discussion on the uses of Hydrogen industrially to manufacture</p>	<p>i) Discussing in groups the daily life uses of hydrogen. ii) Discussing on the relationship</p>	<p>Wall chart, flip charts, pictures showing the production of ammonium fertilizers,</p>	<p>i) Ability to state the uses of hydrogen gas in daily life.</p>	<p>i) Ability to state the uses of hydrogen gas in daily life.</p>	

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							margarine and ammonia. ii) Leading discussion on the relationship between uses and properties of hydrogen.	between the uses and properties of hydrogen.	margarine, uses of hydrogen, properties of hydrogen.		ii) Ability to relate some uses of hydrogen to its properties.	
Competence	General objective	Number of weeks	Week	Main Topic	Sub-topic	Periods	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Ren
Treating and using water while conserving the environment.	To purify and use water while conserving the environment.	2 nd	FEBRUARY	WATER	3.1 Occurrence and nature of water.	2	i) Lead discussion on the occurrence and nature of water. ii) Guiding students to discuss and present the concept of water cycle. iii) Guiding students to draw water cycle and discuss it.	i) Discussing the occurrence and nature of water. ii) Discussing and presenting the concept of water cycle. iii) Drawing and discussing water cycle. iv) Discussing the relationship between water	Wall chart showing occurrence of water, wall chart and picture displaying water cycle.	i) Chemistry for secondary schools, form1&2, Oxford. ii) O-level CHEMISTRY Form 2, BEN.	i) Ability to describe the occurrence and nature of water. ii) Ability to describe water cycle. iii) Ability to relate water cycle and environmental	

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er wi th en vi ro n m en ta l co ns id er at io n.						iv) Guiding students to discuss the relationship between water cycle and environmental conservation.	cycle and environmental conservation.		iii) O-level CHEMISTRY Form 2 Interactive CD, BEN.	conservatio n.
						i) Demonstrating an experiment on boiling and melting points of water. ii) Guiding students to test for water using CoCl_2 paper and anhydrous copper (II) sulphate. iii) Guiding students to carry out reaction between water and some metals. iv) Facilitating discussion on physical and		i) Carrying out an experiment on boiling and melting points of water. ii) Testing for water using CoCl_2 and anhydrous CuSO_4 . iii) Carrying out reactions between water and some metals. iv) Discussing physical and chemical properties of water.	Thermometer, CoCl_2 paper, anhydrous CuSO_4 , water, Na, Ca, Mg, K, litmus paper and wall charts showing physical chemical properties of water.	

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C o m p et e nc e	Gene ral objec tive	N o n t h	W e ek	Mai n Top ic	Sub-topi c	P e ri o d s	chemical properties of water.					
							Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Ren
		F E B R U A R Y	3 rd		3.3 Treatment and purification of water	4	i) Leading students on discussions of different methods of treating and purifying water at home. ii) Visiting large scale water treatment plant. iii) Leading a discussion on different chemicals used to treat large scale of water.	i) Discussing different methods of treating and purifying water at home. ii) Purifying water after boiling by filtering by using a clean piece of cloth. iii) Preparing a simple water filter by using sand, charcoal and gravel and use it for filtering water.	Water, clean piece of cloth, sand, charcoal, gravels, filter paper, boiling vessels (sufurias), water guard pellets/tablet, water treatment plant, wall charts showing large scale.		i) Ability to demonstrate the process of domestic water treatment and purification . ii) Ability to describe the process of urban water treatment. vi) iii) Ability to explain	

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						iv) Lead and summarize discussion on water treatment.	iv) Visiting and writing a report on a large scale water treatment plant. v) Discussing different chemicals used in large scale water treatment and importance of treatment in daily life.			the importance of water treatment and purification .
		4 th		3.4 Uses of water	2	i) Lead and summarizes the discussion on the uses of water in daily life. ii) Guiding students to dissolve different substances in water.	i) Discussing the uses of water in daily life. ii) Dissolving different substances in water and list the dissolved ones.	Wall charts showing uses of water, table salt, sugar, toothpaste, water, kerosene, detergent soap.		i) Ability to state the uses of water in daily life. ii) Ability to compare solubility of different substances in water.

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Competence	General objective	Month	Week	Main Topic	Sub-topic	Periods	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Remarks
Using fuel efficiency and sustainability in using fuels. (ii) To promote the use of fuels with environmental	(i) To importance in efficiency and sustainability in using fuels. (ii) To promote the use of fuels with environmental	N/A	1 st	FUELS AND ENERGY	4.1 Fuel sources	4	i) Leading a discussion of different sources of fuels found in Tanzania.	i) Discussing the different sources of fuels found in Tanzania. ii) Discussing the process of making charcoal.	Kerosene, fire wood, charcoal, petrol, heating gas, diesel, wall charts showing process of making charcoal.	i) Chemistry for secondary schools, form1&2, Oxford. ii) O-level CHEMISTRY Form 2, BEN. iii) O-level CHEMISTRY Form 2	i) Ability to identify different sources of fuels. ii) Ability to describe methods of obtaining fuels from locally available materials. iii) Ability to classify fuels according to their states. ii) Ability to classify	
			2 nd				i) Summarizing the classification of fuels according to their states. ii) Leading students to discuss the	i) Listing fuels according to their states. ii) Discussing the efficiency of different kind of fuels.	Heating gas, charcoal, fire wood, kerosene, piece of wood.	Interactive CD, BEN. iv) You tube(for video clips)		

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ron menta l cons ider at ion.	consi derati on.						efficiency of different kind of fuels. iii) Supervising students to burn different fuels and determine their calorific values.	iii) Burning different fuels and determine their calorific values.				fuels according to efficiency.		
							3 rd	MIDTERM TESTS						
							4 th	MIDTERM BREAK						
							W e e k	Mai n Top ic	Sub-topi c	P e r i o d s	Teaching Activities	Learning Activities	T/L Materials	References
Com pet e nce	Gene ral objec tive	Not h												

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<p>Using fuels efficiently and sustainably in using fuels</p> <p>(ii) To promote the use of fuels with environmental consideration.</p>	<p>(i) To explain the importance in efficiency and sustainability in using fuels</p> <p>(ii) To promote the use of fuels with environmental consideration.</p>	<p>1ST</p> <p>P</p> <p>R</p> <p>L</p>	<p>FUELS AND ENERGY</p>	<p>4.3 Uses of fuels</p>	<p>4</p>	<p>i) Leading a discussion on uses of fuels in daily life and environmental effects of depending on fire wood and charcoal as sources of fuel.</p> <p>ii) Leading a discussion on deforestation, vegetation, and alternative sources of fuels.</p>	<p>i) Discussing the uses of fuels in daily life and environmental effects.</p> <p>ii) Discussing deforestation, vegetation, and alternative sources of fuels.</p>	<p>Wall charts showing uses of fuels, disadvantages of deforestation and alternative sources of fuels.</p>		<p>i) Ability to list uses of fuels.</p> <p>ii) Ability to assess the environmental effect of charcoal and fire wood as sources of fuels.</p>

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ion.	2 nd			4.4 Conservation of energy	4	<p>i) Leading students to discuss the impossibility of destroying or creating energy.</p> <p>ii) Guiding students to perform experiments on the conservation of energy from one form to another and to discuss the results obtained.</p>	<p>i) Discussing the impossibility of destroying or creating energy.(law of conservation of energy)</p> <p>ii) Performing experiments on the conservation of energy from one form to another and to discuss the results.</p>	<p>Wall charts showing energy change, voltaic cell, electric cell, bar magnets, iron fillings, water, source of heat, Cu foil, $H_2SO_4(1M)$, lamp bulb, beaker, Mg ribbon, Abrasive paper, Fe fillings.</p>		<p>i) Ability to explain the law of conservation of energy.</p> <p>ii) Ability to performing experiments on the conservation of energy from one form to another.</p>
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A p pl yi ng pe ri o di cit y to ex pl ai n ch ar ac te ris tic of el e m en ts.	To explai n the struct ure of an atom and perio dic trend N A Y	4 th A P R I L	ATOMI C STRUC TURE	4 th N A Y	5.1 The Atom	2	i) Leading students to discuss Dalton's contribution to the structure of an atom. ii) Leading students to discuss the modern concept of Dalton's atomic structure.	i) Dividing a solid substance into fine indivisible particles and discussing Dalton's contribution to structure of an atom. ii) Analysing the Dalton's atomic theory.	Pieces of chalks, marble chips, pestle, mortar, paper, wall charts showing Dalton's atomic theory, modern periodic table.	i) Chemistry for secondary schools, form1&2, Oxford. ii) O-level CHEMISTRY Form 2, BEN. iii) O-level CHEMISTRY Form 2 Interactive CD, BEN.	i) Ability to explain Dalton's contribution to atomic structure. ii) Ability to explain the modern concept of Dalton's atomic structure.
					5.2 Subatomic particles	2	i) Guiding students to identify the position of sub-atomic particles. ii) Guiding students to discuss the properties of each particle in an atom.	i) Identifying the position of sub-atomic particles. ii) Explaining the properties of each particle in an atom.	Pictures/models of an atom, atomic diagrams. iv) You tube(for video clips)	i) Ability to identify subatomic particles. ii) Ability to explain properties of each sub-atomic particle.	

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			1 st , & 2 nd		5.3 Electronic arrangement	10	<ul style="list-style-type: none"> i) Guiding students to establish the maximum number of electrons in the shell. ii) Guiding students to draw energy shell diagrams of common atoms. iii) Summarizing energy shell diagrams. 	<ul style="list-style-type: none"> i) Establishing the maximum number of electrons in the shell. ii) Drawing energy shell diagrams of common atoms. 	Atomic diagrams, wall charts showing energy shell diagrams.		<ul style="list-style-type: none"> i) Ability to establish the maximum number of electrons in the shells. ii) Ability to draw energy shell diagrams. 	
com pet enc e	Gene ral objec tive	Na t u r a l s	W e e k	Mai n Top ic	Sub-topi c	Pe ri o d s	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Ren

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N A Y	2 nd			5.4 Atomic number, mass number and isotopy.	10	<p>i) Guiding students to discuss the relationship between the atomic number and number of protons and the role played in relating the atomic number and protons.</p> <p>ii) Guiding students on how to obtain mass number.</p> <p>iii) Leading students to discuss the concept of isotopy.</p>	<p>i) Discussing the relationship between the atomic number and number of protons and the role played in relating the atomic number and protons.</p> <p>ii) Obtaining mass number.</p> <p>iii) Discussing the concept of isotopy.</p>	<p>Models/charts, pictures showing the atomic number of elements, playing cards, wall charts showing the number of protons and neutrons of elements, wall charts showing isotopes of C, Cl, O and H.</p>	<p>i) Ability to relate atomic number with number of protons.</p> <p>ii) Ability to the mass number of an atom from number of protons and neutrons.</p> <p>iii) Ability to explain the concept of isotopy.</p>

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			3rd								
				TERMINAL EXAMS							
LONG VACATION											
C o m p et e nc e	Gene ral objec tive	W e ek N o n t h	Main Topic	Sub-top ic	P e ri o d s	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Ren
		5 th M A Y	PERIOD IC CLASSIF ICATIO N	6.1 Periodicity	1	Leading a discussion on the concept of periodicity.	Explaining periodicity.	Wall chart displaying the modern periodic table.	i) Chemistry for secondary schools, form1&2, Oxford.	Ability to explain the concept of periodicity.	
		J U N E 1 st		6.2 General trends	10	i) Guiding students to discuss the change of properties across the period. ii) Guiding students to	i) Listing down the changes of properties across the period. i) Listing down the changes in properties down the group. i) Writing the electronic configurations of 1 st 20 elements.	Modern periodic table and atomic models.	ii) O-level CHEMISTRY Form 2, BEN.	i) Ability to explain changes in properties of elements across the periods	

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						discuss the change of properties down the group. iii) Guiding students to write electronic configurations of 1 st 20 elements.			iii) O-level CHEMISTRY Form 2 Interactive CD, BEN. iv) You tube(for video clips)	and down the groups. ii) Ability to use electronic configurations
	2 nd , 3 rd & 4 th	FORMULA, BONDING AND NOMENCLATURE	7.1 Valence and chemical formulae	12	i) Guiding students to discuss the concept of valency. ii) Using questions and answers in writing simple formula of binary compounds. iii) Leading discussion on the concept of empirical and molecular formula. iv) Summarizing students' activities on	i) Discussing the concept of valency. i) Writing simple formulae of binary compounds. i) Discussing and interpreting the information given the empirical and molecular formula. iv) Calculating the empirical and molecular formulae of various compounds.	Modern periodic table, valence cards, Styrofoam spheres, models/wall charts/pictures showing molecular and empirical formulae.	i) Chemistry for secondary schools, form1&2, Oxford. ii) O-level CHEMISTRY Form 2, BEN. iii) O-level CHEMISTRY Form 2 Interactive CD, BEN.	i) Ability to explain valency. ii) Ability to write simple formulae of binary compounds. iii) Ability to explain and calculate the empirical and molecular formulae.	

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Competence	General objective	Number of weeks	Week	Main Topic	Sub-topic	Periods	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	iv) YouTube (for video clips)	Resources
			1st		7.2 Oxidation states	4	i) Leading discussion and summarizing the concepts of oxidation states. ii) Providing activities on valence and oxidation states.	i) Discussing the concept of oxidation states. ii) Performing activities on valence and oxidation states.	Modern periodic table, wall charts showing oxidation state and valence.		Ability to explain the oxidation states and to differentiate oxidation state from valence.		
			2 nd		7.3 Radicals	4	i) Leading questions on names and formulae of radicals. ii) Illustrating the writing of the chemical formulae of	i) Practicing writing and naming formulae of common radicals. ii) Practicing writing of chemical formulae of common compounds.	Modern periodic table, wall charts showing common radicals.		i) Ability to explain the concept of radicals. ii) Ability to write chemical formulae of compounds.		

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							common compounds.					
			3 rd									
MIDTERM BREAK												
Competence	General objective	Number of weeks	Week	Main Topic	Sub-topic	Periods	Teaching Activities	Learning Activities	T/L Materials	References	Assessment	Remarks
		S E P T E M B	1 st & 2 nd		7.4 Covalent bonding	8	i) Guiding the discussion on the concept of covalent bonding. ii) Guiding students to draw electron diagram to show	i) Discussing the concept of covalent bonding. ii) Drawing electron diagram to show covalent bonding in binary molecules.	Modern periodic table, wall charts showing covalent compounds, O ₂ , H ₂ , kerosene,		i) Ability to explain the covalent bonding. ii) Ability to state the properties of	

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		E R					covalent bonding in binary molecules. iii) Guiding students to discuss the properties of covalent compounds. iv) Guiding students to perform experiments on covalent compounds.	iii) Discussing the properties of covalent compounds. iv) Performing experiments on covalent compounds.	diesel, water, bulb, connecting wires, carbon electrodes and beaker.		covalent bonding.
		3 rd & 4 th		7.5 Electrovalent bonding	8	i) ii) iii)	Leading discussion on electrovalent bonding. Guiding students to draw electro diagrams to show electrovalent bonding. Guiding students to discuss the properties of electrovalent compounds.	i) Discussing the concept of electrovalent compounds. ii) Drawing electron diagrams to ionic bonding. iii) Discussing the properties of electrovalent compounds. iv) Performing simple experiments on ionic bonding.	Modern periodic table, wall charts showing electrovalent compounds, table salt, H ₂ O, KCl, bulb, ammeter, connecting wires, carbon electrodes, and beakers.		i) Ability to explain the ionic bonding. ii) Ability to state the properties of ionic compounds.

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							iv) Guiding students to perform experiments on ionic compounds.					
FORM 2 NATIONAL EXAMINATIONS												