



## ОСНОВНО-КИСЛОТНОЕ ВЗАИМОДЕЙСТВИЕ

*основание + кис. амф. гидроксид = соль + вода*

- $\text{LiOH} + \text{H}_2\text{SO}_4 =$
- $\text{LiOH} + \text{Al}(\text{OH})_3 =$
- $\text{Be}(\text{OH})_2 + \text{HNO}_3 =$
- $\text{Be}(\text{OH})_2 + \text{Zn}(\text{OH})_2 =$
- $\text{NaOH} + \text{H}_2\text{Cr}_2\text{O}_7 =$
- $\text{NaOH} + \text{Fe}(\text{OH})_3 =$
- $\text{Mg}(\text{OH})_2 + \text{H}_3\text{AsO}_4 =$
- $\text{Mg}(\text{OH})_2 + \text{Be}(\text{OH})_2 =$
- $\text{Al}(\text{OH})_3 + \text{H}_2\text{CrO}_4 =$
- $\text{Al}(\text{OH})_3 + \text{Zn}(\text{OH})_2 =$
- $\text{Zn}(\text{OH})_2 + \text{H}_2\text{SeO}_4 =$
- $\text{Zn}(\text{OH})_2 + \text{HClO}_4 =$
- $\text{Cr}(\text{OH})_2 + \text{HNO}_3 =$
- $\text{Cr}(\text{OH})_2 + \text{HBrO}_3 =$
- $\text{Fe}(\text{OH})_2 + \text{HBr} =$
- $\text{Fe}(\text{OH})_2 + \text{HClO}_2 =$
- $\text{Fe}(\text{OH})_3 + \text{HCl} =$
- $\text{CsOH} + \text{H}_2\text{CO}_3 =$
- $\text{CsOH} + \text{Be}(\text{OH})_2 =$
- $\text{Ba}(\text{OH})_2 + \text{H}_2\text{SO}_3 =$

