

Storyline Unit Design

Understanding by Design (UbD) Template*

Unit	Engines of the Past to Engines of the Future	Course(s)	Regents Chemistry
Designed by	Valley Central Chemistry	Time Frame	6 Weeks
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Anchor Model

*UbD Unit Planner is from Wiggins, Grant and McTighe, Jay. Understanding by Design Guide to Creating High-Quality Units. Alexandria, VA: Association for Supervision and Curriculum Development. 2011.

Stage 1: Desired Results

Performance Expectations

HS-PS1-2: Simple Chemical Reactions

Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

HS-PS1-4: Total Bond Energy Change in Chemical Reactions

Develop a model to illustrate that the release or absorption of energy from a chemical reaction system depends upon the changes in total bond energy. (Energy and Matter)

HS-PS1-5: Collision Theory and Rates of Reaction

Apply scientific principles and evidence to provide an explanation about the effects of changing the temperature or concentration of the reacting particles on the rate at which a reaction occurs. (Patterns)

HS-PS3-1: Energy Change in Components of a System

Create a computational model to calculate the change in the energy of one component in a system when the change in energy of the other component(s) and energy flows in and out of the system are known. (Systems and System Models)

Anchoring Phenomenon

[Anchoring Phenomenon Worksheet](#)

Enduring Understandings

Type Here

Essential Questions

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Stage 2: Assessments

HS-PS1-2	Predicting Products	Assessment	Key	Evidence Statement
HS-PS1-4	Chemical Reaction Energy Transfer	Assessment	Key	Evidence Statement
HS-PS1-5	Alka Seltzer Rockets	Assessment	Key	Evidence Statement
HS-PS3-1	Algodoo Ball Bounce	Assessment	Key	Evidence Statement

[Assessment Screening Tools](#)

Backward Design Elements

What new skills (practices) will students need to learn?	What thinking concepts will students need to learn?	What science concepts will students need to learn?
1-3 <i>Analyzing graphs and data</i> <i>Model energy changes into and out of a system</i> <i>Interpreting a data table</i> <i>Calculate Thermal Energy of a system</i> 1-2, 1-3 & 1-5 <i>Construct and revise an explanations based on reliable evidence</i> <i>Assume theories and laws are the same today as in the past and continue in the future</i> 1-4 <i>Labeling bonds bonds on a diagram</i> 1-5 <i>Model Particle Diagrams</i> 1-5 <i>Provide an explanation of phenomena using evidence and solve design problems, taking into account possible unanticipated effects.</i>	1-3 <i>Systems/System Models: Stability & Change</i> 1-2 & 1-5 <i>Observe different patterns in different scales to provide evidence of phenomena</i> 1-4 <i>Energy & Matter (Energy)</i>	1-4 <i>Energy absorbed as bonds broken Energy released as bonds formed</i> 1-4 <i>Electrons can be transferred or shared to form a chemical bond</i> 1-4 <i>Rearranging atoms to form new molecules results in a change in total bond energy in the system that will affect the kinetic energy of the surroundings</i> 1-2 Periodic table shows repeating patterns according to the outer level electrons 1-2 Atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions. 1-5 <i>Chemical processes, their rates, and whether or not energy is stored or released can be understood in terms of the collisions of molecules and the rearrangements of atoms into new molecules,</i>



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Stage 3: Learning Plan

 Phenomenon or Problem	 Learning Performance - What will they do? The three dimensions woven together into a single learning performance.	 Why is this important? How does this activity help build understanding of the anchoring phenomenon.	 Learning Experience - How will they do it? Graphic organizers, protocols, scaffolds, labs, mini-lesson, student discourse, etc.
<i>Type Here</i>	<i>Type Here</i>	<i>Type Here</i>	<i>Type Here</i>
Formative Assessment - What information are you collecting to know that they met the target?		Assess students	
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Summative Assessment What information are you collecting to know that they met the target?			

Materials / Resources

Vocabulary

HS-PS1-2

Chemical reaction (e.g. sodium and chloride, carbon and oxygen, carbon and hydrogen)
 Reactants and products
 Bonds (i.e. ionic, covalent)
 Reaction type (e.g. formation of ionic compound, combustion of hydrocarbons)
 Electronegativity
 Main group elements
 Valence electrons
 Periodic table
 Chemical properties
 Patterns

HS-PS1-5

Temperature
 Concentration
 Chemical reaction
 Reactants and products
 Chemical bonds
 Molecules
 Rate of reaction
 Collision theory
 Kinetic energy
 Patterns

HS-PS1-4

Chemical reaction system
 - Chemicals
 - Surroundings
 Total bond energy
 Bonds
 Potential energy
 - Bond energy of reactants
 - Bond energy of products
 Kinetic energy
 - Surroundings by molecular collisions

HS-PS3-1

System and boundaries
 Components
 Energy (thermal energy, kinetic energy, and/or the energies in gravitational, magnetic, or electric fields)
 Energy flow

Mini Lessons

[System Level 5 - Simulating Systems Mini-Lesson](#)
[System Level 5 - Simulating Systems Thinking Slides](#)
[Patterns Level 7 - Causal Patterns at Varying Scale](#)
[Patterns Level 7 - Causal Patterns at Varying Scale](#)
[Matter and Energy Level 5 - Cycles and Flows](#)
[Matter and Energy Level 5 - Cycles and Flows Thinking Slides](#)

Graphic Organizers

[Phenomena Observation Graphic Organizer](#)



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[Questioning Graphic Organizer](#)
[Modeling Graphic Organizer](#)
[Planning an Investigation Organizer - Experimental](#)
[Planning an Investigation Organizer - Observational](#)
[Investigation Evidence Organizer](#)
[Engaging in Argumentation Organizer](#)

Differentiation / Modifications



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◀ **HS-PS1-2 - Simple Chemical Reactions**◀ **HS-PS1-4 - Total Bond Energy Change in Chemical Reactions**◀ **HS-PS1-5 - Collision Theory and Rates of Reaction**◀ **HS-PS3-1 - Energy Change in Components of a System**◀ **Local and Relevant**◀ **Favorite**

- Combustion engine in a vehicle and development of the battery ▶
 - Gasoline/Fuel
- Baking soda and vinegar volcano ▶▶
- Hot/Cold Pack??? ▶f
- Road salt/Brines
- Electrolysis of water
- Digestions & Body Systems?
- Photosynthesis and plant growth
- Cooking an egg
- Baking cookies with varying amounts of ingredients
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<https://static1.squarespace.com/static/56ef1da37da24f301fccaacd/t/58d485e5ebbd1a1fe2025acc/1490322976040/HSCChemSkeleton-1.0v.pdf>

Screening Tools

Back to [Stage 2](#)

HS-PS1-2: Simple Chemical Reactions

Predicting Products

[Assessment](#)[Key](#)[Evidence Statement](#)

Reflections: Type Here			
	No	Partial	Yes
1. The assessment contains a phenomenon (science) or a problem (engineering)			
2. The prompts match the Science and Engineering Practice (SEP) and engage students in sense making.			
3. The stimuli have multiple and sufficient information needed to utilize the SEP . (e.g. multiple data sets to analyze)			
4. The prompts elicit observable understanding of the Disciplinary Core Idea (DCI) .			
5. The prompts explicitly mention the Crosscutting Concept (CCC) .			
6. The prompts include language (i.e. bullets) from grade appropriate progressions. (SEP) (DCI) (CCC)			
7. The graphic organizers provide space for the observable features (e.g. 1, 2, 3...) in the			

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evidence statement. (e.g. claim, evidence and reasoning)			
8. The entire assessment contains information that is scientifically accurate and properly attributed. (e.g. don't make up data and include the source)			
9. The prompts point in the direction of explaining a phenomenon (science) or designing a solution (engineering).			
10. The phenomenon or problem is authentic, interesting, and requires students to figure something out.			
11. The phenomenon or problem is novel to show the transfer of knowledge. (i.e. not in the unit)			



HS-PS1-4: Total Bond Energy Change in Chemical Reactions

Chemical Reaction Energy Transfer

[Assessment](#)

Key

[Evidence Statement](#)Reflections: [Type Here](#)

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HS-PS1-5: Collision Theory and Rates of Reaction

Alka Seltzer Rockets

[Assessment](#)

Key

[Evidence Statement](#)

Reflections: Type Here			
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HS-PS3-1: Energy Change in Components of a System

Algodo Ball Bounce

[Assessment](#)[Key](#)[Evidence Statement](#)Reflections: [Type Here](#)

	No	Partial	Yes
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