

## Polyatomic Ions

Ammonium	$(\text{NH}_4)^{+1}$
Acetate	$(\text{C}_2\text{H}_3\text{O}_2)^{-1}$
Bicarbonate	$(\text{HCO}_3)^{-1}$
Carbonate	$(\text{CO}_3)^{-2}$
Chlorite	$(\text{ClO}_2)^{-1}$
Chlorate	$(\text{ClO}_3)^{-1}$
Cyanide	$(\text{CN})^{-1}$
Chromate	$(\text{CrO}_4)^{-2}$
Dichromate	$(\text{Cr}_2\text{O}_7)^{-2}$
Hydronium	$(\text{H}_3\text{O})^{+1}$
Hydroxide	$(\text{OH})^{-1}$
Nitrite	$(\text{NO}_2)^{-1}$
Nitrate	$(\text{NO}_3)^{-1}$
Phosphite	$(\text{PO}_3)^{-3}$
Phosphate	$(\text{PO}_4)^{-3}$
Sulfite	$(\text{SO}_3)^{-2}$
Sulfate	$(\text{SO}_4)^{-2}$