

## ***MODULE 3 VOCABULARY METRIC - PART 2***

**Este documento no es para que lo conserve. Necesita permanecer en este salón de clases.**

1. Physical Property: A characteristic of a pure substance that can be observed without changing it into another substance
2. Boiling Point: The temperature at which a liquid changes to a gas
3. Chemical Property: the ability of a substance to undergo a specific chemical change
4. Combustion: the process of burning something
5. Dissolving: a physical change in which particles of a substance are separated by water particles
6. Flammability: A material's ability to burn in the presence of oxygen
7. Melting Point: The temperature at which a solid becomes a liquid
8. Orbital (atomic): allowed paths around the nucleus which the electron can circle
9. Precipitate: A solid that forms from a solution during a chemical reaction.
10. Reactivity: The ease and speed with which an element combines, or reacts, with other elements and compounds.
11. Solubility: the ability of one substance to dissolve in another at a given temperature and pressure
12. Valence Electron: Electrons on the outermost energy level of an atom