

Name: \_\_\_\_\_

**Reaction Type and Balancing Practice**

Balance the following chemical equations, AND designate the reaction type:

Chemical Equation	Reaction Type
$\text{___PbCl}_2 + \text{___AgNO}_3 \square \text{___Pb(NO}_3)_2 + \text{___AgCl}$	
$\text{___NH}_3 + \text{___HCl} \square \text{___NH}_4\text{Cl}$	
$\text{___Al}_2\text{S}_3 \square \text{___Al} + \text{___S}_8$	
$\text{___AlCl}_3 + \text{___Na}_2\text{SO}_4 \square \text{___Al}_2(\text{SO}_4)_3 + \text{___NaCl}$	
$\text{___Fe} + \text{___O}_2 \square \text{___Fe}_2\text{O}_3$	
$\text{___C}_4\text{H}_{12} + \text{___O}_2 \square \text{___CO}_2 + \text{___H}_2\text{O}$	
$\text{___Na} + \text{___CaF}_2 \square \text{___Ca} + \text{___NaF}$	
$\text{___C}_3\text{H}_7 + \text{___O}_2 \square \text{___CO}_2 + \text{___H}_2\text{O}$	
$\text{___Li}_2\text{SO}_4 + \text{___MgCl}_2 \square \text{___LiCl} + \text{___MgSO}_4$	
$\text{___MgO} \square \text{___Mg} + \text{___O}_2$	
$\text{___NaOH} + \text{___CuSO}_4 \square \text{___Na}_2\text{SO}_4 + \text{___Cu(OH)}_2$	
$\text{___Ca}_3\text{P} + \text{___Cl}_2 \square \text{___CaCl}_2 + \text{___P}$	
$\text{___H}_2\text{SO}_4 + \text{___Fe} \square \text{___H}_2 + \text{___FeSO}_4$	
$\text{___Ba}_3(\text{PO}_3)_2 + \text{___Fe(NO}_3)_4 \square \text{___Fe}_3(\text{PO}_3)_4 + \text{___Ba(NO}_3)_2$	
$\text{___C}_4\text{H}_9 + \text{___O}_2 \square \text{___CO}_2 + \text{___H}_2\text{O}$	

