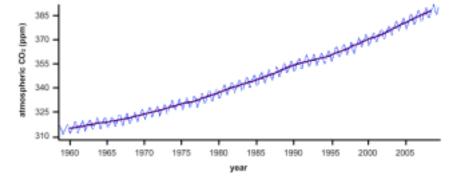
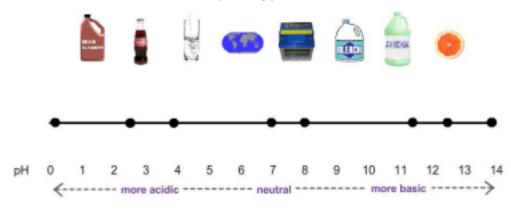
## SEP: Analyzing & Interpreting Data SEP: Developing and Using Models

- Go to: <a href="https://depts.washington.edu/vurchin/">https://depts.washington.edu/vurchin/</a>
- Click on "Our Acidifying Ocean" in the list on the left under Ecology and Environment.
- 1. *Click on Intro*: How much of the Earth is covered in oceans?
- 2. *Click on Air*: What is the main idea of the graph to the right?

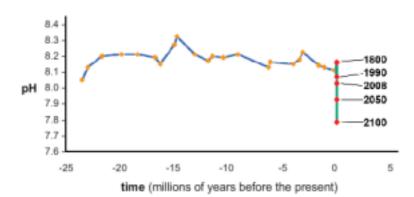


- 3. What is the trend shown in the graph to the right?
- 4. What is the difference between the solid line and the wavy line in the graph above?
- 5. Does this data represent an ecosystem in homeostasis or not? Explain how you know this.
- 6. *Click on pH*: Connect the items to their corresponding pH level below.



- 7. At what pH should ocean water be?
- 8. Which of the pH readings surprised you most? \_\_\_\_\_ Why?
- 9. What does pH actually measure? (You may need to look this up on your own)

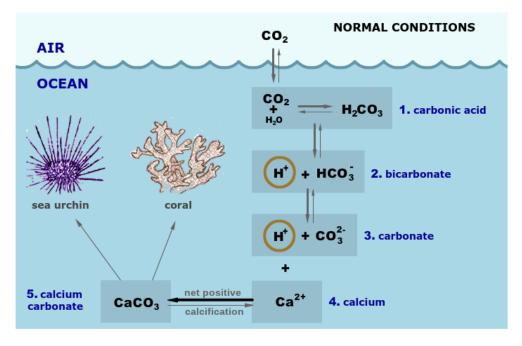
- 10. Where would you think most of your body exists on the pH scale?
- 11. Where in your body would you expect a very low pH?
- 12. *Click on Ocean pH:* What is the main idea of the graph to the right?



- 13. What is the trend shown in the graph to the right?
- 14. What is the difference between the horizontal line and the vertical line in the graph above?
- 15. Does the data, in the graph above, represent an ecosystem in homeostasis or not? Explain how you know this.

Click on Chemistry: Examine the diagram and read the text in the box below the diagram to answer the questions.

- 16. What are shells and skeletons of sea creatures made of?
- 17. Why do the arrows go in both directions?
- 18. Why are some arrows thick and others are not?
- 19. Describe the conversion from atmospheric CO2 to calcium carbonate under normal conditions.



- 20. Hover over the picture to make it show red arrows. Draw the red arrows on the diagram above to show acidified conditions.
- 21. Why do the red arrows seem thicker than the black arrows?
- 22. What molecules increase due to a higher concentration of atmospheric CO2?
- 23. What does an increase in H+ ion concentration do to the pH value?
- 24. What does an increase in H+ ion concentration do to the acidity of the water?
- 25. Which molecule decreases due to a higher concentration of atmospheric CO2?
- 26. What impact does higher atmospheric CO2 have on urchin shell development?