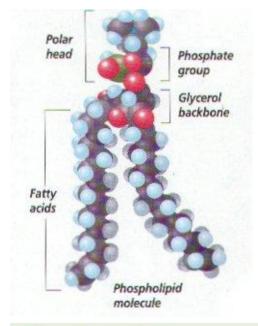
Biology 12 Biological Molecules III Lipids

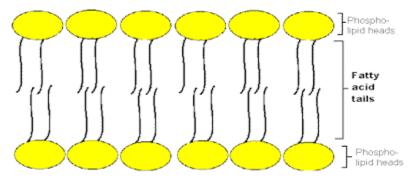
Lipids are usually not soluble in	Familiar lipids include
(for example lard and butter) which are	at room temperature.
Another group of familiar lipids are the	(for example corn oil and
canola oil). These are at roon	n temperature. Fats have a number
of functions in the body. List three of them b	pelow.
1.	
2.	
3.	
A fat or oil forms when three	react with one
molecule. The synthesis reaction is shown be	elow.
Glycerol has three groups (-	OH) which allow the alveered to
	,
react with three	to form one triglyceride (fat)
and three molecules of	
Why is a triglyceride also called a NEUTRAL F	FAT?
	Charles .
Fatty acids are a chain with	the acidic group -
on the end. Most fatty acids conta	ain 16 or 18 carbon 🏻 🎇 🌉 🥂 🥂
atoms, but smaller ones do exist.	
What is the difference between saturated and	d unsaturated fatty
acids? Choose from the diagrams below and	explain the difference.

Molecule	Saturated or	Distinguishing
	Unsaturated	characteristic and
		examples

TO T			
0 H H H H H H H H H H H H H H H H H H H			
Emulsifiers			
Soaps are formed from a	plus ar	n	
A molecule of soap has a polar end and a non polar end. Explain			
how soap emulsifies fats and oi	ils.		
Why would a person without a	gallbladder have trouble di	igesting fatty foods?	
Phospholipids			
What possible groups take the place of a third fatty acid in a phospholipids?			
The following diagrams show different representations of a phospholipids.			
Explain why a phospholipids are not electrically neutral.			



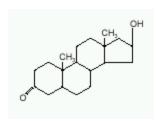
Phospholipids make up the cell membrane. The polarity of the phospholipids makes a bilayer possible. Label the polar and nonpolar regions of the bilayer.



Steroids and Cholesterol

Steroid molecules all have a backbone of
They differ according to arrangement of atoms in the rings and
the type of
Cholesterol is the precursor to several steroids. Give their functions.

STEROID	FUNCTION
Aldosterone	
Estrogen/Testerone	



Lipids are organic compounds t	hat are	in wate	r. In the body,
they serve as	term energy st	orage molecules.	Lipids include
fats,, and _			
The 3 most important classes o	f lipids are neutra	al fats,	, and
Oil, fat, butter are all composed	of lipid molecule	es called	(or
). Neut	tral fats are compo	osed of two
types of molecules:	and		
Most fatty acids contain about _		_ carbon atoms in	a long chain.

Saturated fatty acids have no	bonds between carbon atoms,
and tend to be solid at room temperature.	Unsaturated fatty acids are most
often found in vegetable oils, and account	for the fact that they are liquid at
room temperature.	
Butter contains a large proportion of	fatty acids. Excess
intake of this type of fatty acid is known to	cause attacks and
strokes.	
Soap is a formed when	a
is reacted with an in	organic base such as
Soap allows oils to be	mixed with water by
the oils.	
A phospholipid is a lipid made of glycero	l, 2 fatty acids, and a phosphate
group. It is the primary component of me	mbranes. The phosphate "head" is
, the tail is	
are small lipids containi	ng rings that are all derived from
cholesterol. An important function of these	e compounds are sex
like progesterone.	

LIPIDS

- 1. Lipids are a wide variety of compounds, more frequently known by their **common names**, including *fats*, *oils*, *waxes*.
- 2. are all **insoluble** in water.
- 3. Functions of Lipids include the following:
- 1. **Padding** of vital organs
- 2. **Insulation** ("blubber")
- 3. Long-Term **Energy storage** (fat is excellent for storing energy in the least amount of space, and packs 9.1 calories of energy per gram, versus 4.4 for carbohydrates and proteins).
- 4. **Structural** (e.g. **cell membranes**, white matter of brain etc.)
- 4. **chemical messengers** (e.g. steroid hormones, prostaglandins).

There are four main types:

- i. **FATTY ACIDS**: a long chain of carbons with hydrogens attached, ending in an acid group (-COOH). There are two main types:
 - 5. **saturated fatty acids no double bonds** between carbons. All carbons are "saturated" with hydrogens. **Solid** at room temp. These are the "bad" dietary fats (e.g. butter, lard, meat fat), which are known to contribute to heart disease and strokes.
- 6. <u>unsaturated fatty acids</u> have one (monounsaturated) or more (polyunsaturated) double

bonds between carbons in chain. **Liquid** at room temperature. e.g. vegetable oils, Omega-3 unsaturated fatty acids. Are thought to be better for your heart than saturated fats.

- ii. **NEUTRAL FATS**: (also called **Triglycerides**)
- 7. formed by dehydration synthesis rx. between **glycerol** (a molecule of 3 hydrated carbons and **3 fatty acids**.
- 8. non-charged, non-polar molecules. Do not mix with water unless a soap is added.

Neutral fats are sometimes drawn like this:

- 9. **Soap** is made by combining a **base** and a **fatty acid**.
- 10. soaps are **polar**, will mix with water. Soap molecules surround oil droplets to their polar ends project outwards, causing the oil to **disperse** in water (this process called **EMULSIFICATION**).
- iii. PHOSPHOLIPIDS: important components of cell membranes
- 11. same basic structure as neutral fats **except** that 1 fatty acid is replaced by a phosphate group with a charged nitrogen attached.
- 12. phospholipids have a Phosphate-containing "head" and two long fatty acid tails. Head is hvdrophilic ("water-loving"), tail is hvdrophobic ("water-fearing")
- iv. **STEROIDS**: a different type of lipid
- 13. They are multi-ringed structures, all derived from **CHOLESTEROL**
- 14. You've heard many bad things about cholesterol, but it is actually an **essential molecule** found in every cell in your body (it forms parts of cell membranes, for example).
- 15. The problem is that <u>dietary</u> cholesterol helps to form <u>arterial plaques</u>, which lead to <u>strokes</u> and <u>heart attacks</u>. Dietary cholesterol only found in <u>animal products</u> (meat, fish, poultry, dairy products). There is <u>no cholesterol in plant foods</u>. Your blood cholesterol should be no more than 150 mg/dl
- 16. Steroids can function as **chemical messengers**, and form many important **hormones** (e.g. **testosterone**, **estrogen**, **aldosterone**, **cortisol**) that have a wide variety of affects on cells, tissues, and organs (especially sex characteristics, ion balance, and gluconeogenesis).