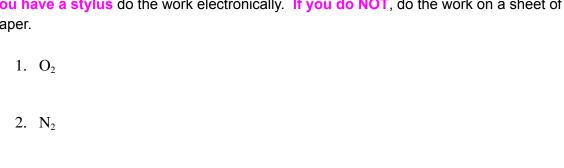
Lewis Dot Structures Practice Multiple Bonds ONLY

Draw a Lewis Dot Structure for each molecule below. All of these structures require MULTIPLE (double and triple) bonds.

Some of these compounds can form multiple isomers. Identify which compounds can have multiple correct Lewis Dot Structures, and draw 2 different isomers for each of them!

I recommend doing this work on a separate document so you have the needed space. If you have a stylus do the work electronically. If you do NOT, do the work on a sheet of paper.





3. C_2H_4





$$8. \quad C_4H_4Cl_2$$

10.
$$C_5H_{10}O_3$$

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Some of these compounds can form multiple isomers Identify which compounds can have multiple correct Lewis Dot Structures, and draw 2 different isomers for each of

1. O₂ :
$$\ddot{G} = \ddot{O}$$
:

6.
$$N_2H_2 \longrightarrow H - \ddot{D} = D - 1$$

★ 8.
$$C_4H_4Cl_2$$
 + H_4 | $C_4 = C_4 = C_4 - C_4 = C_4$