

Name:

Osmosis in Gummy Bears

(20 points)

Diffusion is the movement of molecules an area of *high concentration to an area of lower concentration*. When molecules do that it's called moving down a concentration gradient. To visualize diffusion, think about placing a few drops of green food coloring in water. At first most of the water is clear with a small amount of dark green food coloring concentrated in the center where the drops are placed, but over a few minutes the molecules of food coloring spread out in the water until they are evenly distributed among the water molecules and the entire glass appears a light green color. The same thing happens in the air when someone sprays perfume inside a room. At first the molecules of perfume are concentrated and strong-smelling in a small area, but over time they diffuse through the air in the room until they are evenly distributed and the entire room smells weakly of perfume.

Osmosis is the diffusion of water through a membrane. During osmosis *water moves from an area of higher concentration of water molecules to an area of lower concentration of water molecules*. Often, whether inside a living organism or in a laboratory experiment, water contains other molecules, or solutes. The higher the concentration of solutes, the lower the concentration of water, so *water will move from low solute concentration to high solute concentration*.

1. Define the terms: **(2 points)**

Diffusion:

Osmosis:

2. In diffusion and osmosis, in which direction are particles moving (from high concentration to low concentration or from low concentration to high concentration)? **(1 point)**

3. Define the following terms: **(3 points)**

Hypertonic solution:

Hypotonic solution:

Isotonic solution:

Procedure: Day 1

- Each group should take three gummy bears. Use the ruler to measure and record the starting length (in millimeters) of all three gummy bears. Record these measurements in Table 1
- Weigh each gummy bear and record the mass (in milligrams) in Table 1.
- Gather three 100 mL beakers and label them #1, #2, and #3.
- In beaker #1, add one inch of regular water and place one of the gummy bears inside.
- In beaker #2, add one inch of salt water, as well as one gummy bear.
- In beaker #3 add nothing except the third gummy bear.
- Cover all three glasses with plastic and leave overnight.

Table 1: Gummy Bear Measurements Before Soaking

Gummy Bear Measurements	#1	#2	#3
Starting length (mm)			
Starting Mass (mg)			

4. Predict what you think will happen to each of the gummy bears. **(3 points)**

Procedure: Day 2

- Measure the final length of each gummy bear, and calculate the ending length in Table 2.
- Determine the change in length (final length – starting length) of each gummy bear and record your results.
- Measure the final mass of each gummy bear and record the mass in Table 2.
- Determine the change in mass (final mass – starting mass) of each gummy bear and record your results.

Table 2: Gummy Bear Measurements After Soaking (3 points)

Gummy Bear Measurements	#1	#2	#3
Final length (mm)			
Change in Length (mm)			
Final Mass (mg)			
Change in Mass (mg)			

5. Using your table for reference, describe what happened to Gummy Bear #1, #2 and #3. **(1 point)**

6. Did your observations match your prediction? If no, explain what you think accounts for the difference. **(2 points)**

7. a) What kind of solution (hypertonic, hypotonic, or isotonic) was the tap water? How do you know? **(1 point)**

b) Was water moving into Gummy Bear #1 or out of the gummy bear? _____ **(1 point)**

c) What is this process called? _____ **(1 point)**

8. a) What kind of solution (hypertonic, hypotonic, or isotonic) was the salt water? How do you know? **(1 point)**

b) What water moving into Gummy Bear #2 or out of the gummy bear? _____ **(1 point)**

c) What is this process called? _____ **(1 point)**

9. Like Gummy Bears, cells are also permeable to water. How can you predict whether water will move into a cell or out of a cell? **(2 points)**