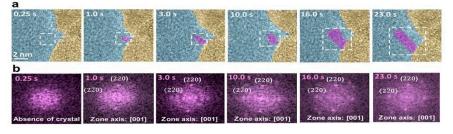
Real-time, nanoscale observation of ice nucleation on TiO2 nanoparticles in water

Abhijit Phakatkar, Reza Shahbazian, Constantine Megaridis, <u>Tolou Shokuhfar</u> University of Illinois at Chicago, Chicago, IL, 60607

The understanding of heterogenous ice nucleation at water-solid interfaces is a subject of considerable interest owing to its extreme importance in geological, biological, and atmospheric sciences. 1-4 The existing experimental methods limit the temporal and spatial resolutions of ice crystals nucleation at nanointerfaces. Herein, we propose the in-situ graphene liquid cell (GLC) transmission electron microscopy (TEM) approach to investigate the real time heterogenous ice nucleation phenomenon at cryogenic temperature. The present study evaluates the heterogenous ice crystals nucleation at the nanointerface of water and anatase titanium dioxide (TiO₂) nanoparticles using cold stage TEM holder. Atomic resolution in-situ liquid TEM results showed the nucleation and growth of cubic phase ice crystals at the facets of anatase TiO₂ nanoparticle. Fast Fourier transform (FFT) analysis confirmed the predominant presence of characteristic (220) lattice planes of cubic phase ice crystals affiliated to 2.2 Å d-spacings. The observations of formation of cubic phase of ice crystals at TiO₂ nanointerface are in lined with the findings reported by Souda et. al⁵, evaluated using reflection high energy electron diffraction (RHEED) characterization technique. The obtained atomic resolution in-situ liquid TEM results also provide insights on electron beam induced melting events of ice crystals. Cubic phase ice crystals were relatively stable at the TiO₂- water nanointerface in comparison with those in the bulk liquid, suggesting the possible influence of radical scavenging capability of TiO₂ nanoparticles mitigating the electron beam byproducts. Results also indicate that the absence of ice crystals around the gold nanoparticles likely attributed with their inherent hydrophobic surface properties. Moreover, the key dynamic events of cubic phase single ice crystal nucleation and growth and the ice crystal growth via multiple crystallites attachments along $(\overline{220})$ active facets were evaluated in detail. Briefly, the present work demonstrates the capability of in-situ GLC TEM route to study the heterogenous ice nucleation and growth dynamic events at



water-metal oxide nanointerface.

Figure 1. In-situ liquid TEM real time heterogenous ice nucleation at water-TiO₂ nanoparticle interface at cryogenic temperature. (a) TEM micrographs indicating temporal study of single ice crystal nucleation and growth events. (TiO₂ nanoparticles, ice crystal, and

water are highlighted by gold color, purple color, and cyan color, respectively.) (b) Fast Fourier transform (FFT) analysis confirming the cubic phase of ice crystal.

References:

- 1. Cober, S. G.; Isaac, G. A.; Strapp, J. W., Characterizations of aircraft icing environments that include supercooled large drops. Journal of Applied Meteorology **2001**, 40 (11), 1984-2002.
- 2. DeMott, P. J.; Rogers, D. C.; Kreidenweis, S. M., The susceptibility of ice formation in upper tropospheric clouds to insoluble aerosol components. Journal of Geophysical Research: Atmospheres 1997, 102 (D16), 19575-19584.
- 3. Knight, C. A.; De Vries, A. L.; Oolman, L. D., Fish antifreeze protein and the freezing and recrystallization of ice. Nature 1984, 308 (5956), 295-296.
- 4. Morris, C. E.; Georgakopoulos, D. G.; Sands, D. C. In Ice nucleation active bacteria and their potential role in precipitation, Journal de Physique IV (Proceedings), EDP sciences: 2004; pp 87-103.

5. Souda, R.; Aizawa, T.; Sugiyama, N.; Takeguchi, M., Nucleation and growth of water ice on oxide surfaces: the influence of a precursor to water dissociation. Physical Chemistry Chemical Physics 2020, 22 (36), 20515-20523.