## IONIC COMPOUNDS and FORMULAS WORKSHEET

NAME:

Fill in the boxes below with the correct formula for each ionic compound.

	Cl <sup>-1</sup>	Br <sup>-1</sup>	O <sup>-2</sup>	N <sup>-3</sup>	NO <sub>3</sub> -1	$SO_4^{-2}$	CO <sub>3</sub> -2	MnO <sub>4</sub> -1
Na <sup>+1</sup>								
Li <sup>+1</sup>								
Ca <sup>+2</sup>								
Al <sup>+3</sup>								
K <sup>+1</sup>								
Cu <sup>+1</sup>								
Cu <sup>+2</sup>								
Fe <sup>+2</sup>								
Fe <sup>+3</sup>								
Pb <sup>+2</sup>								
Pb <sup>+4</sup>								
Sn <sup>+2</sup>								
Sn <sup>+4</sup>								
Zn <sup>+2</sup>								
NH <sub>4</sub> <sup>+1</sup>								