Springdale Public Schools Chemistry-Integrated Unit 4



| Title of Unit | Periodicity | Grade Level | secondary |
|--------------------------|----------------------|---------------|-----------|
| Subject | Chemistry-Integrated | Time Frame | 4 weeks |
| Developed By SPS Science | | Date Modified | 7/20/2021 |

Identify Desired Results

Standards Covered in this **Unit**

(Evidence statements are LINKED from the standard number. Evidence statements assist in clarifying outcomes)

<u>CI-PS1-1</u> Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level of atoms. [AR Clarification Statement: This PE is fully addressed in this course. Examples of properties that could be predicted from patterns could include reactivity of metals, types of bonds formed, numbers of bonds formed, and reactions with oxygen.] [AR Assessment

<u>CI-ESS1-4</u> Use mathematical or computational representations to predict the motion of orbiting objects in the solar

Boundary: Assessment is limited to main group elements. Assessment does not include exceptions to periodic trends.]

system.(3.c)** [Clarification Statement: Emphasis is on Newtonian gravitational laws governing orbital motions, which apply to human-made satellites as well as planets and moons.] [Assessment Boundary: Mathematical representations for the gravitational attraction of bodies and Kepler's laws of orbital motions should not deal with more than two bodies, nor involve calculus.]

<u>CI-PS1-2</u> Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.* [AR Clarification Statement: This PE is fully addressed in this course. Examples of chemical reactions could include the reaction of sodium and chlorine, carbon and oxygen, and carbon and hydrogen.] [Assessment Boundary: Assessment is limited to chemical reactions involving main group elements and combustion reactions.]

<u>CI-PS2-4</u> Use mathematical representations of Newton's Law of Gravitation and Coulomb's Law to describe and predict the gravitational and electrostatic forces between objects. [Clarification Statement: Emphasis is on both quantitative and conceptual descriptions of gravitational and electric fields.] [Assessment Boundary: Assessment is limited to systems with two objects.]

| Learning Outcomes for the Unit What relevant goals will this <u>unit</u> address? These <u>must come</u> from the standards. | | | | |
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| Key Vocabulary for the U | Jnit | | | |
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| Periodic | Energy level | Chemical property | Physical property | Electrostatic force |
| Period | Group | Electronegative | Coulomb's Law | Coulombic attraction |
| Valence | | | | |
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| Enduring Understanding | gs for the Unit (for discus | sion within science and ac | ross content areas) | |
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| Essential Questions for the Unit | | | | |
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| Misunderstandings That Will Be Addressed | | | | |
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| Content Literacy Skills for the Unit (Interpretation of data, experimental design, SEPs, CCs) [minimum list] | | | | |
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Assessment Evidence

What type(s) of Common Formative Assessment (CFA) will be given?

| What type of District Formative Assessment will be given? |
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TBD

| Overview of <u>All</u> Choices of Lessons for Unit This is not a lesson plan!. | | | |
|--|---|-----------------------------|--|
| Standard(s) # | Formative Assessment(s) (Indicate which is the CFA) | Main Instructional Strategy | Activity/Activities |
| | graph analysis | Interactive | Tuva Lab - Elements in the Periodic Table (free) |
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| Common Resources | | |
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| Title and Description of Usage | Location | |
| This is an option for a lab notebook that is virtual. | Blank Digital Lab Notebook | |
| This is a lesson plan template that pairs with the Unit plan. It allows for easy information transfer. | <u>Lesson Plan Template</u> | |
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