## Molecular Modeling Practice

## **Objective:**

To practice using the molecular model kits as a tool for determining the molecular shape and polarity of a covalent compound.

## **Procedure:**

- 1. Draw a Lewis Dot Structure for each molecule in the table.
- 2. Predict the shape using VSEPR Theory.
- 3. Build a model using the model kit. If needed, <u>here is an electronic model kit</u> that can be used to visualize compound shape if the physical kit is not available.
- 4. Use the model to predict polarity. Look for symmetry.

## **Molecules:**

$H_2$	$H_2CO$
HBr	$C_2H_2$
$H_2O$	CH <sub>3</sub> Cl
$PH_3$	НСООН
CH <sub>4</sub>	HCN
HClO	$H_2O_2$
$N_2$	

Make a table like the one below:

Formula	LDS	VSEPR Formula	VSEPR Shape (draw)	Polar or Nonpolar