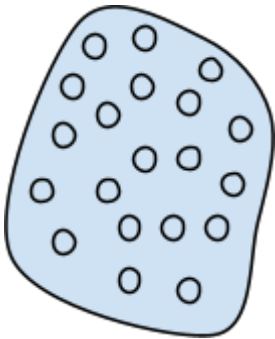
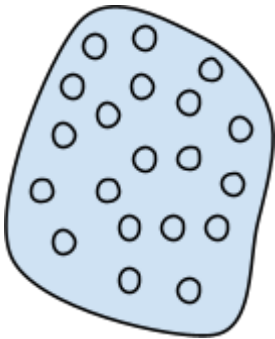
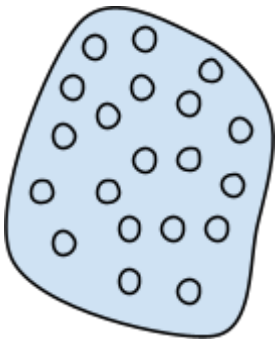
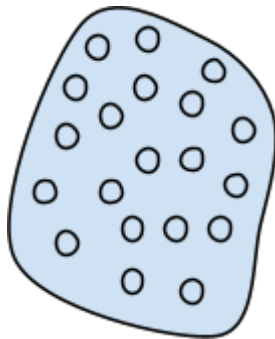


## Rocks and Absolute Age

*Half-life is the time it takes for half of the radioactive atoms, parent element, to decay into the non-radioactive element, the daughter element.*

The rock below contains Thomasium and the **half-life is 5,000 years**.

The total amount of atoms it contains is 20.

|   |   |
|---|---|
| <p><b>0 Half-lives</b> (Ratio 20:0)</p>  <p>Age = 0 yrs</p>            | <p><b>1 Half-life</b> (Ratio ____:____)</p>  <p>Age = _____ yrs</p>    |
| <p><b>2 Half-lives</b> (Ratio ____:____)</p>  <p>Age = _____ yrs</p> | <p><b>3 Half-lives</b> (Ratio ____:____)</p>  <p>Age = _____ yrs</p> |

1. Draw in the rock the ratio of radioactive atoms to non-radioactive atoms.

- pick one color for the radioactive element
- pick a different color for the non-radioactive element
- make a key to your colors on the right

☐ \_\_\_\_\_  
☐ \_\_\_\_\_

2. Figure out the ratio after each half-life.

3. Color the correct number of atoms for that ratio.

4. Calculate age, write answer on line under the rock.

5. *What would the age of the rock be after 4 half-lives?* \_\_\_\_\_

a. *What would the ratio be?* \_\_\_\_\_