SUNRISE ONE EXAM END TERM 1 2021 F1 CHEMISTRY MARKING SCHEME

- 1. *i*) Study of the structure, properties, and composition of matter and the changes that matter undergoes. $\sqrt{1}$
 - (ii) Manufacture of drugs to fight diseases $\sqrt{1}$ Manufacture of fertilizers to increase food production $\sqrt{1}$ Manufacture of alternative fabrics like nylon and polyester. $\sqrt{1}$ Manufacture of soap
- 2. (i) Any substance that alters the body function. $\sqrt{1}$
 - (ii) Tobacco√1 Alcohol√1

Khat√1

Bhang

(iii) Hallucination√1 Depression√1

Addiction

3. (a). Burette $\sqrt{1}$

Pipette√1

Volumetric flask

Syringe

(b). Glass is used for heating. $\sqrt{1}$

Glass does not react with reagents $\sqrt{1}$ Any two

Glass is easy to clean

Glass is transparent

- 4. A mass of burning gas√1
- 5. Don't eat in the labaratory√1

Any three

Don't run in the laboratory√1

Always wait for teachers intruction $\sqrt{1}$

Open doors and window for fresh air

- 6. (i) A component of two or more subtances that are not chemically combined $\sqrt{1}$
- (ii). Filtering $\sqrt{1}$ Decanting $\sqrt{1}$

7

Chemical change	Physical change	
(i) Not easily reversible	Are reversible√1	
(ii) Always accompanied by heat change	No heat change√1	
(iii) Reactants have different masses	No change in mass√1	
compared to products formed		
(iv) Always produce new substance	No new substances formed√1	

- 8. a) Sublimation1mk
 - b) Chromatography1mk
 - c) Fractional Distillation**1mk**
 - d) Solvent extraction 1mk
 - e) Filtration/Decantation1mk
- 9. a) Non luminous $\sqrt{1}$
 - b) A pale blue region $\sqrt{1}$
 - B Green -blue region $\sqrt{1}$
 - b) $A\sqrt{1}$, it contains completely burnt gases. $\sqrt{1}$

10.

Luminous		Non-luminous	
i.	Produced when air hole is closed.	i.	Produced when air hole is open. $\sqrt{1}$
ii.	Used for lighting.	ii.	Used for heating. $\sqrt{1}$
iii.	Has 4 parts.	iii.	Has 3 parts. $\sqrt{1}$
iv.	Produces soot.	iv.	Doesn't produce soot. √
V.	Fairly hot	V.	Very hot√1

- 11. a) Separating funnel. $\sqrt{1}$
 - b)Water is denser than kerosene. $\sqrt{1}$
 - c) Water and petrol. √1
 Water and oil
 Water and diesel

 Any one
- 12. a) Chemical $\sqrt{1}$
 - b) Physical $\sqrt{1}$
 - c) Physical√1
 - d) Chemical √1
 - e) Physical √1
- 13. i) A: Thermometer $\sqrt{1}$

- B: Fractionating column $\sqrt{1}$
- C: Lie big condenser√1
- ii) Distillate√1
- iii) A to measure the temperature at which the liquids boil. $\sqrt{1}$ B condense the vapour back into the flask before the boiling point is reached. $\sqrt{1}$
- iv) Separating components of crude oil. $\sqrt{1//}$ Obtaining oxygen and nitrogen gases from liquefied air.
- 14. Mortar: √1 grinding solid substances √1Desiccator: √1 Keeping substances free of moisture √1
- 15. Liquids√1, Solid√1, Gas√1
- 16. Chemistry teacher√1/Medicine√1/Beautician
- 17. a) A: Melting $\sqrt{1}$ D: Condensation $\sqrt{1}$ B: Freezing $\sqrt{1}$ E: Sublimation $\sqrt{1}$ C: Evaporation $\sqrt{1}$ F: deposition $\sqrt{1}$
 - b) Alluminium Chloride 1/Iron (III) Chloride 1/iodine/Benzoic acid
- 18. Add water to the mixture and $stir\sqrt{1}$ Filter to obtain sand as the residue and salt solution as the filtrate $\sqrt{1}$ Evaporate the filtrate to obtain $salt\sqrt{1}$
- 19. Diagram