# Module 5.2: Electronegativity and Polarized Bonds

# Learning Objectives

Upon successful completion of this sub-module, students will be able to:

- LO 5.2.1 Explain the polarization of electron density caused by electronegativity differences of atoms in a covalent bond.
- LO 5.2.2 Define partial charges on atoms.
- LO 5.2.3 Understand how to represent bond dipole moments with crossed arrow notation.
- LO 5.2.4 Predict bond moment magnitude using the Pauling Electronegativity scale.
- LO 5.2.5 Describe the type of bond between two atoms: non-polar covalent, polar covalent or ionic.

# Prior Knowledge Review

• Module 5.1: Lewis Structures of Simple Molecular Compounds

## Instructional Materials

#### **Textbook Sections**

- OpenStax Chemistry Atoms First 2e
  - o 7.6 Molecular Structure and Polarity (LO 5.2.1-5.2.5)
- Libretext A Molecular Approach (Tro)
  - 9.6: Electronegativity and Bond Polarity (LO 5.2.1-5.2.5)

#### **Videos Resources**

• Electronegativity and polar covalent bonds (LO 5.2.1-5.2.4)

- Description: This video clip (0:00-4:47) shows how to classify different types of bonds using differences in electronegativity and how to show a dipole moment using a vector pointing to the more electronegative atoms.
- License: By Patreon.com/TheScienceClassroom used under a standard Youtube license.
- Polar or non-polar covalent bonds (LO 5.2.1-5.2.5)
  - Description: This video clip (0:00-10:24) shows how to classify different types of bonds using differences in electronegativity and how to show a dipole moment using a vector pointing to the more electronegative atoms. This video has more examples than the one above.
  - License: By Sciencepost used under a standard Youtube license.
- Polarizing power and covalent bonds (LO 5.2.4)
  - Description: This video clip (0:24-2:35) discusses polarizability based on ionic size and number of electrons.
  - License: By Dr. Amanda Brindley at the University of California Irvine Open Education Consortium under Creative Commons Attribution license BY-NC-SA.

#### Other Resources

- Calculator: Chemical Bond Polarity Calculator (LO 5.2.1-5.2.5)
  - Description: This website calculates the difference in electronegativity between any two atoms.
  - License: By Calculators.live. All rights reserved but open to use by students and instructors.

### **Activities and Assessments**

## Study Guides / Worksheets

- Worksheet: <u>Lewis Structures and Bonding</u> (LO 5.2.1-5.2.5)
  - Description: This worksheet includes problems requiring calculations of differences in electronegativity to determine the type of bond.

- Teaching Tip: This worksheet includes problems from earlier and later material will likely need to be edited. Appropriate problems start at problem #12.
- Answer Key: <u>Lewis Structures and Bonding Key</u> (LO 5.2.1-5.2.5)
- License: By Dr. Christine Yerkes and the Department of Chemistry at the University of Illinois Urbana-Champaign (<u>Dr. Yerkes' website</u>) used with permission. Please attribute upon use.

## **Textbook Exercises**

- Openstax Atom First 2e Chapter 4 (LO 5.2.1-5.2.5)
  - Description: These end-of-chapter exercises cover all the topics in the chapter and include several exercises.
  - Teaching Tip: Solutions are not provided for the even-numbered exercises which might be used in assessment.
  - o Access: On-line only (Neither downloadable nor editable).
  - Answer Key: The odd-numbered exercises are provided. The direct link to the solutions does not work but one can click on any odd-numbered questions to get to the solutions page.
  - License: By Openstax at Rice University and used under Creative Commons Attribution License 4.0.