

Module Descriptions

A **module** is a self-contained **learning unit** within a higher education program that includes thematically related courses and is assigned a **fixed number of credits**. It follows specific **learning objectives**, includes an **assessment component**, and contributes to achieving the qualifications of a degree program. In some countries, "modules" are also named "courses".

Please provide a module description for each module. In addition to the compulsory and elective modules, this also includes credited internships and the final thesis.

Please summarize all module descriptions in one document (Module Handbook) and create a table of contents so that the modules can be found easily.

Module designation	Inorganic Structural Chemistry	
Semester(s) in which the module is taught	1st Semester	
Person responsible for the module	Prof. Dr. Hari Sutrisno, M.Si	
Language	Indonesian	
Relation to curriculum	Compulsory / elective / specialisation	
Teaching methods	Lecture, Discussion, Demonstration, Experimentation, Independent Study, Project Work, Fieldwork	
Workload (incl. contact hours, self-study hours)	Contact hours: 2 x 50 minutes per week Independent study and assignments: Estimated 3–4 hours/week Total ECTS equivalent: 2 credits (approx. 85–90 hours total workload)	
Credit points	2 credits (SKS)	
Required and recommended prerequisites for joining the module	None	
Module objectives/intended learning outcomes	Inorganic Structurale Chemistry courses are courses for students of Master of Education in Chemistry with descriptions including: chemical structure description, symmetry and molecular groups, chemical bonds and lattice energy, molecular structures 1 (compounds of the main group elements) and 2 (transition metal compounds), crystal gratings, symmetry and groups crystals, X-ray diffraction instruments and determination of simple crystal structures. This course aims to enable students to understand the structure and grid contained in molecular compounds 1 and 2.	



Content	Subjects include: Description of chemical structure Theory of repulsion of valence electron pairs Symmetry and molecular groups Chemical bonds and lattice energy Atom size size Symmetry and crystal groups Molecular structure 1: compounds of the main group elements Molecular structure 2: transition metal compounds Structure of nonmetal elements X-ray diffractometer Determination of simple crystal structure	
Examination forms	Midterm Test, Final Exam, Assignments, Project-Based Learning and Presentations, Case Studies	
Study and examination requirements	Minimum attendance: 75% of lectures and full participation in fieldwork Evaluation Components:	
	Assessment Type	Weight (%)
	Attendance	5
	Assignments	10
	Midterm Exam	15
	Final Exam	20
	Case Studies	15
	Team-based Projects	35
	Total	100
Reading list	 Hari Sutrisno, Dyah Purwaningsing, A.K. Prodjosantosa. (2023). Fundamental Kristalokimia. UNY Press Muller, U., (2006). Inorganic Structural Chemistry, second edition. West Sussex: John Wiley & Sons Ltd Huheey, J. E., Keiter, E. A. & Keiter, R. L. (1993). Inorganic Chemistry: Principle of Structure and Reactivity. New York: Harper Collins College Publisher. Li, W. K., Zhou, G. D. & Wai Mak, T. C. (2008). Advanced Structural Inorganic Chemistry. New York: Oxford Science Publication Miessler, G. L. & Tarr, D. A. (2009). Inorganic Chemistry, third edition. New Delhi: Pearson Education West, A. R. (1989). Solid State Chemistry and Its Applications. Singapore: John Wiley & Sons Ltd. S2. Journal Inorganic Chemistry 	



Prepared by	Verified by:	Authorized by:
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