

# Y9: 3A The structure of the atom

## Skill sharpener: Processing and analysing

### Skills achievement standards

**Year 8:** Select and construct appropriate representations to organise and process data and information.

**Year 9:** Select and construct appropriate representations to organise, process and summarise data and information.

**Year 10:** Select and construct effective representations to organise, process and summarise data and information.

### Representing the atom

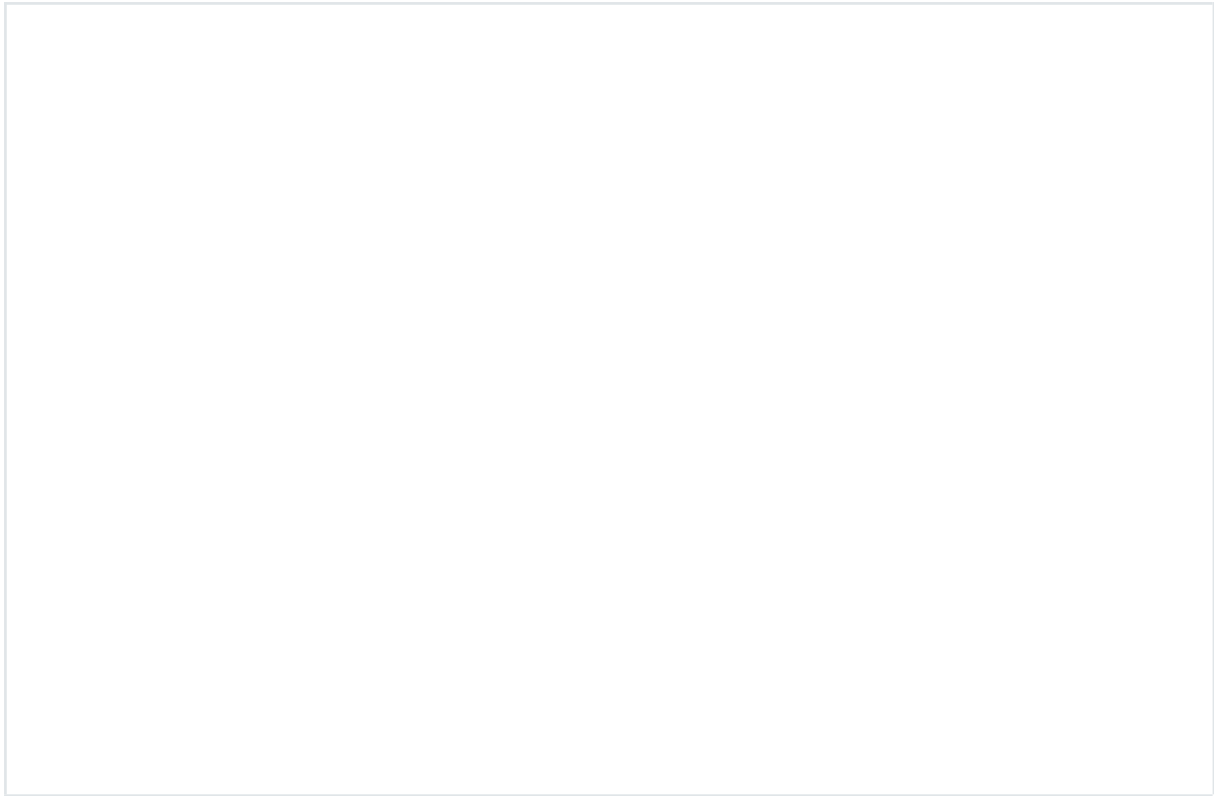
1. Choose an element from the first 20 and design a model using everyday objects.

<b>1</b> <b>H</b> 1.0 hydrogen					<b>2</b> <b>He</b> 4.0 helium		
<b>3</b> <b>Li</b> 6.9 lithium	<b>4</b> <b>Be</b> 9.0 beryllium						
<b>11</b> <b>Na</b> 23.0 sodium	<b>12</b> <b>Mg</b> 24.3 magnesium	<b>5</b> <b>B</b> 10.8 boron	<b>6</b> <b>C</b> 12.0 carbon	<b>7</b> <b>N</b> 14.0 nitrogen	<b>8</b> <b>O</b> 16.0 oxygen	<b>9</b> <b>F</b> 19.0 fluorine	<b>10</b> <b>Ne</b> 20.2 neon
<b>19</b> <b>K</b> 39.1 potassium	<b>20</b> <b>Ca</b> 40.1 calcium	<b>13</b> <b>Al</b> 27.0 aluminium	<b>14</b> <b>Si</b> 28.1 silicon	<b>15</b> <b>P</b> 31.0 phosphorus	<b>16</b> <b>S</b> 32.1 sulfur	<b>17</b> <b>Cl</b> 35.5 chlorine	<b>18</b> <b>Ar</b> 39.9 argon

Include in your model:

- the subatomic particles – protons, neutrons, and electrons
- the location of each subatomic particle
- the position of electrons on each shell.

2. Submit a labelled drawing of your model.



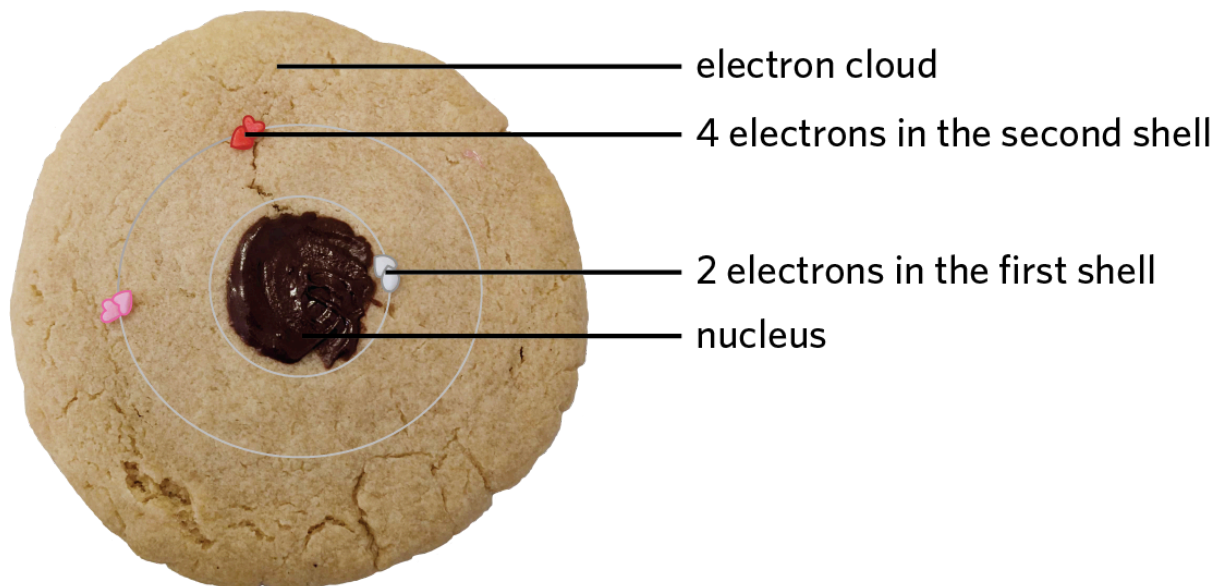
### Extension

Analyse the model you have developed. Come up with a list of advantages and disadvantages of the model when considering all the information and models known about the atom.

Advantages	Disadvantages
<ul style="list-style-type: none"><li>•</li></ul>	<ul style="list-style-type: none"><li>•</li></ul>

## Example answer

### Representing a Carbon atom



### Extension

Advantages	Disadvantages
<ul style="list-style-type: none"><li>• shows the electron cloud as an area where the electrons exist</li><li>• nucleus is placed in the middle</li><li>• displays electrons on different levels based on scattered placement</li></ul>	<ul style="list-style-type: none"><li>• no display of the protons and neutrons in the nucleus – can't count to make sure that there are the correct number of each subatomic particle</li><li>• not in motion so the electron movement in the cloud is not modelled</li><li>• no electron orbits shown either – making it difficult to check the right number of electrons are placed based on 2, 8, 8, 18 rule</li></ul>