

	Formula	Cation symbol	Anion symbol	Ionic compound name
1	VCl_4	V^{4+}	Cl^-	Vanadium (IV) chloride
2	BaF_2			
3	LiOH			
4	CuSO_4			
5	Na_2CO_3			
6	$\text{Ca}_3(\text{PO}_4)_2$			
7	$(\text{NH}_4)_2\text{S}$			
8	NH_4NO_3			
9	Ag_2SO_3			
10	$\text{Pb}(\text{OH})_4$			
11	$\text{Nb}(\text{CN})_3$			
12	$\text{Zn}(\text{NO}_2)_2$			
13	Cu_2O			
14	CuO			
15	HgCl			
16	HgCl_2			
17	FeN			
18	Fe_3N_2			
19	$\text{Sn}_3(\text{PO}_4)_4$			
20	PbCrO_4			
21	$\text{Pb}(\text{Cr}_2\text{O}_7)_2$			
22	$\text{Sb}(\text{ClO})_3 \bullet 4\text{H}_2\text{O}$			
23	$\text{Sb}_2(\text{SO}_4)_5 \bullet 6\text{H}_2\text{O}$			
24	$\text{CoPO}_4 \bullet 3\text{H}_2\text{O}$			
25	$\text{Co}(\text{ClO}_4)_2 \bullet 10\text{H}_2\text{O}$			
26	$\text{Au}_2\text{Cr}_2\text{O}_7$			
27	$(\text{NH}_4)_3\text{PO}_3 \bullet 3\text{H}_2\text{O}$			
28	$\text{Pd}(\text{SCN})_4 \bullet 5\text{H}_2\text{O}$			
29	$(\text{NH}_4)_2\text{HPO}_3 \bullet \text{H}_2\text{O}$			