PS2.C: Stability and Instability in Physical Systems

Why are some physical systems more stable than others?

Events and processes in a system typically involve multiple interactions occurring simultaneously or in seguence. The system's stability and its rate of evolution depend on the balance or imbalance among these multiple effects.

A stable system is one in which the internal and external forces are such that any small change leading to forces that tend to increase that change (e.g., a ball at the top of a hill). A system can be changing but have a stable repeating cycle of changes, with regular patterns of change that allow predictions about the system's future (e.g., Earth orbiting the sun). And a stable system can appear to be unchanging when flows or processes within it are going on at opposite but equal rates (e.g., water in a dam at a constant height but with water flowing in that offsets the water flowing out; a person maintaining steady weight but eating food, burning calories, and excreting waste).

Stability and instability in any system depend on the balance of competing effects. A steady state of a complex system can be maintained through a set of feedback mechanisms, but changes in conditions can move the system out of its range of stability (e.g., homeostasis breaks down at too high or too low a temperature). With no energy inputs, a system starting out in an unstable state will continue to change until it reaches a stable configuration (e.g., the temperatures of hot and cold objects in contact). Viewed at a given scale, stable systems may appear static or dynamic. Conditions and properties of the objects within a system affect the rates of energy transfer and thus how fast or slowly a process occurs (e.g., heat conduction, the diffusion of particles in a fluid).

When a system has a great number of component pieces, one may not be able to predict much about its precise future. For such systems (e.g., with very many colliding molecules), one can often predict average but not detailed properties and behaviors (e.g., average temperature, motion, and rates of chemical change but not the trajectories of particular molecules).

Grade Band End Points (from NRC Framework)

| K-2 | 3-5 | MS | HS |
|--|---|--|--|
| Whether an object stays still or moves often depends on the | A system can change as it moves in one direction (e.g., a ball | A stable system is one in which any small change results in | Systems often change in predictable ways; understanding the |
| effects of multiple pushes and pulls on it (e.g., multiple players | rolling down a hill), shifts back and forth (e.g., a swinging | forces that return the system to its prior state (e.g., a weight | forces that drive the transformations and cycles within a system, |
| trying to pull an object in different directions). It is useful to | pendulum), or goes through cyclical patterns (e.g., day and | hanging from a string). A system can be static but unstable (e.g., | as well as the forces imposed on the system from the outside, |
| investigate what pushes and pulls keep something in place (e.g., | night). Examining how the forces on and within the system | a pencil standing on end). A system can be changing but have a | helps predict its behavior under a variety of conditions. |
| a ball on a slope, a ladder leaning on a wall) as well as what | change as it moves can help to explain the system's patterns of | stable repeating cycle of changes; such observed regular | |
| makes something change or move. | change. | patterns allow predictions about the system's future (e.g., Earth | When a system has a great number of component pieces, one |
| | | orbiting the sun). Many systems, both natural and engineered, | may not be able to predict much about its precise future. For |
| | A system can appear to be unchanging when processes within | rely on feedback mechanisms to maintain stability, but they can | such systems (e.g., with very many colliding molecules), one |
| | the system are occurring at opposite but equal rates (e.g., water | function only within a limited range of conditions. With no energy | can often predict average but not detailed properties and |
| | behind a dam is at a constant height because water is flowing in | | |
| | at the same rate that water is flowing out). Changes can happen | change until it reaches a stable configuration (e.g., sand in an | chemical change but not the trajectories or other changes of |
| | very quickly or very slowly and are sometimes hard to see (e.g., | hourglass). | particular molecules). Systems may evolve in unpredictable |
| | plant growth). Conditions and properties of the objects within a | | ways when the outcome depends sensitively on the starting |
| | system affect how fast or slowly a process occurs (e.g., heat | | condition and the starting condition cannot be specified precisely |
| | conduction rates). | | enough to distinguish between different possible outcomes. |

Note: The Grade Band Endpoints have been identified from the A Framework for K-12 Science Education (2012).

Performance Expectations

| | K-2 | 3-5 | MS | HS |
|---|------|------|------|------|
| ١ | N/A. | N/A. | N/A. | N/A. |

DCI Elements

| K-2 | 3-5 | MS | HS |
|------|------|------|------|
| N/A. | N/A. | N/A. | N/A. |

Note: The DCI Elements come from the bulleted statements of NGSS Performance Expectations.

Sample Assessment Targets

| K-2 | 3-5 | MS | HS |
|------|------|------|------|
| I/A. | N/A. | N/A. | N/A. |

Note: The Assessment Targets listed above have been generated from CAST Item Specification documents and the Evidence Statements.