

Neutralization Reaction

<https://www.aplustopper.com/meant-neutralization-reaction/>

<https://tinyurl.com/y8dr3o93>

<https://goo.gl/ONgy1J>

<http://bit.ly/2FzLkkg>

<http://tiny.cc/v28qqy>

<http://cutt.us/DB0C>

<https://sites.google.com/site/aplustoppertnotes/neutralization-reaction>

<https://goo.gl/jHqdce>

<http://cbsetuts.blogspot.in/2018/02/neutralization-reaction.html>

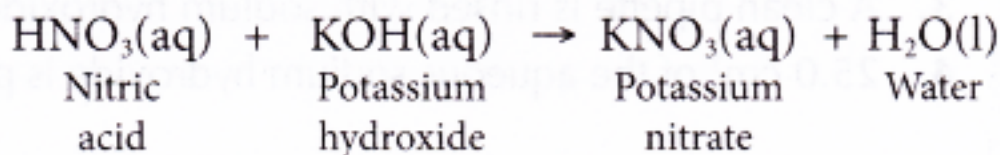
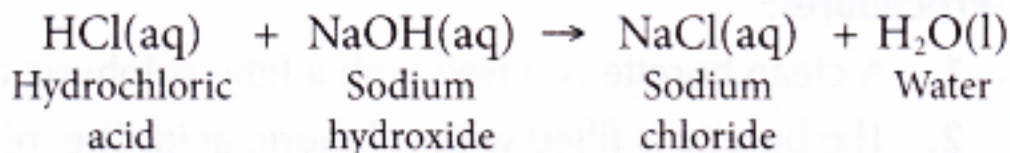
<https://goo.gl/P55N4Y>

<https://goo.gl/FA7h79>

<https://goo.gl/hEU9Nh>

The reaction of an acid and a base is called a **neutralisation reaction**. In this reaction, the acidity of an acid is neutralised by an alkali. At the same time, the alkalinity of the alkali is neutralised by the acid. A salt and water are the only products of a neutralisation reaction.

Acid + alkali → salt + water



Read more about [Neutralization Reaction](#)